

Status on SDT simulation

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Calculation (density/Mass fraction)

Case) gas-A: 80%, gas-B:20%
 density : ρ_A (g/cm³), ρ_B (g/cm³)
 molecular weight : N_A, N_B

Density of gas mixture :

$$\rho_{AB} = (N_A * 0.80 + N_B * 0.20) / 22.4 / 1000 \text{ [g/cm}^3\text{]}$$

a check) C₄H₁₀ only, N=58.12 (from a list)

$$\rho = 58.12 / 22.4 / 1000 = 2.595E-3 \text{ [g/cm}^3\text{]}$$

1mol = 22.4L (0° C) →

$$1\text{mol} = 22.4 * (273+20) / 273 = 24.04 \text{ L}$$

$$\rho = 58.12 / 24 / 1000 = 2.422E-3 \text{ [g/cm}^3\text{]}$$

Atomic and nuclear properties of butane (C₄H₁₀)

Quantity	Value	Units	Value	Units
<Z/A>	0.59497			
Specific gravity (20° C, 1 atm)	2.489E-03	g cm ⁻³		
Mean excitation energy	48.3	eV		
Minimum ionization	2.278	MeV g ⁻¹ cm ²	5.670E-03	MeV cm ⁻¹
Nuclear collision length	55.5	g cm ⁻²	2.231E+04	cm
Nuclear interaction length	77.1	g cm ⁻²	3.098E+04	cm
Pion collision length	83.2	g cm ⁻²	3.342E+04	cm
Pion interaction length	109.0	g cm ⁻²	4.377E+04	cm
Radiation length	45.23	g cm ⁻²	1.817E+04	cm
Critical energy	131.26	MeV (for e ⁻)	128.44	MeV (for e ⁺)
Molière radius	7.31	g cm ⁻²	2935.	cm
Plasma energy $\hbar\omega_p$	1.11	eV		
Muon critical energy	1557.	GeV		
Melting point	134.9	K	-138.2	C
Boiling point @ 1 atm	272.6	K	-0.5000	C

Composition:

Elem	Z	Atomic frac*	Mass frac
H	1	10.00	0.173408
C	6	4.00	0.826592

* calculated from mass fraction data

a bit different . Though it is enough for current study ...

Example 1) He: 90%, C₄H₁₀ : 10% , N_{He} = 4.003, N_{C₄H₁₀} = 58.12

$$\rho = (4.003 \cdot 0.90 + 58.12 \cdot 0.10) / 24 / 1000 = \underline{3.923E-4} \text{ [g/cm}^3\text{]}$$

mass fraction :

$$\begin{aligned} \text{He (90\%)} : \text{C}_4(10\%) : \text{H}_{10}(10\%) &= 4.003 \cdot 0.90 : \\ 12.0107 \cdot 4 \cdot 0.10 : 1.008 \cdot 10 \cdot 0.10 &= 3.6027 : 4.80428 : 1.008 \\ &= \underline{0.38265 : 0.51028 : 0.10706} \end{aligned}$$



comparison with previous settings which was done using rough numbers (e.g. H=1) and referring FCC examples

```
<!-- He : isoC4H10 = 90% : 10% -->
<material name="GasHe_90Isob_10">
  <D value="0.00039839999999999999" unit="g/cm3" />
  <fraction n="0.3826351004462046" ref="He"/>
  <fraction n="0.106371937371285891" ref="H" />
  <fraction n="0.5109929621825095" ref="C" />
</material>
```

again, the precision might be enough . . .

	Atomic mass [g/mole]
H	1.008
C	12.0107
O	15.999
He	4.0026
Ar	39.948

Next

- A trial , He (50%) : isoC₄H₁₀ (50%)
- Make a simple guide(shell scripts?) for my current procedure to reproduce the plots regarding dE/dx study (today ?)
- From a discussion that try to reproduce the IP/momentum resolution plots, done at LDTv2.0, with the CEPCSW
 - confirmation of the LDT code/materials
 - check the CEPCSW framework (possibly with the latest version) , to know how we can achieve it

