

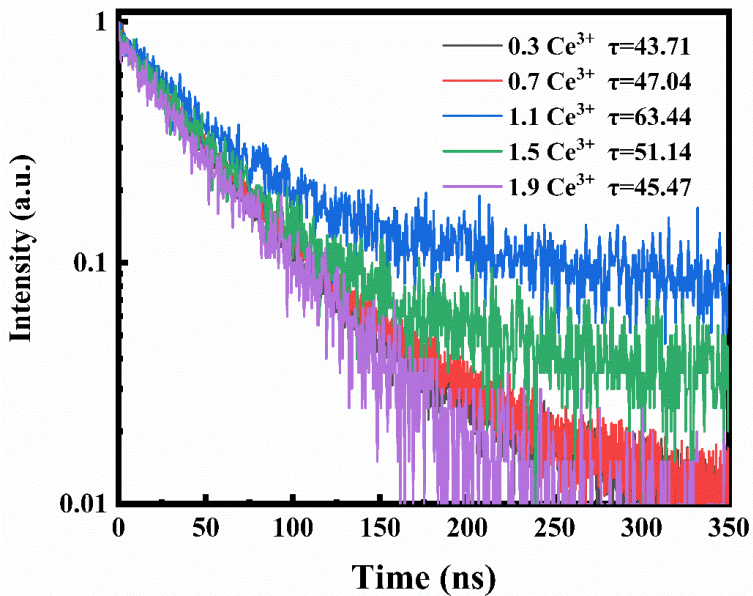
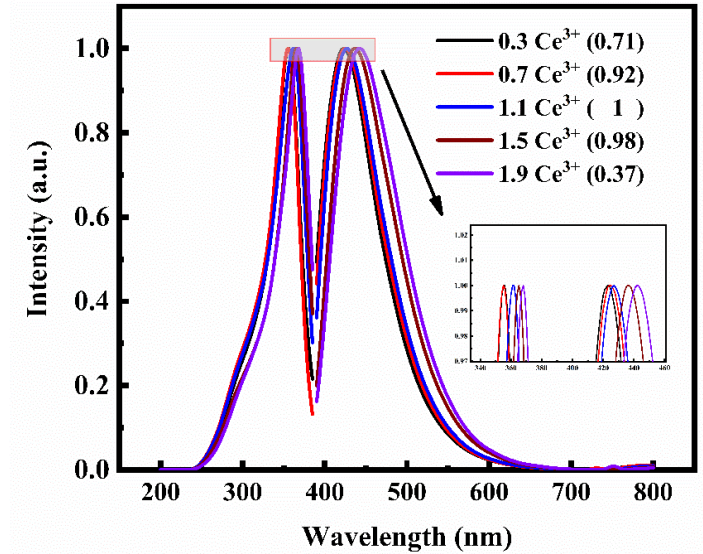
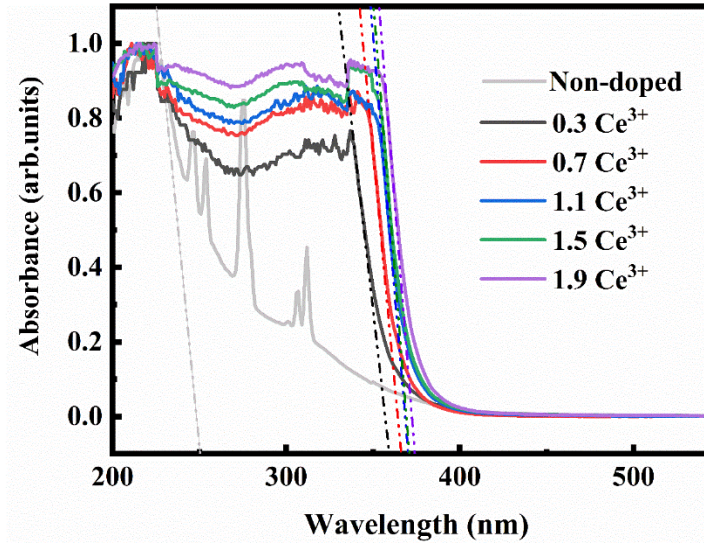
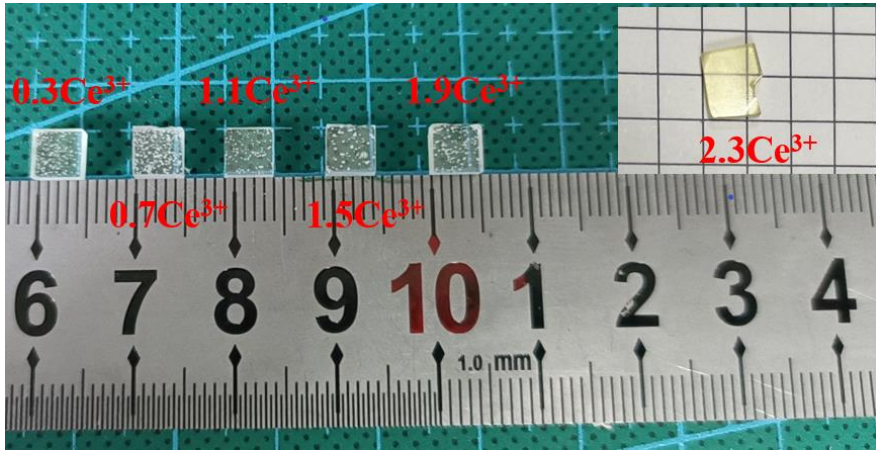


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# Enhanced photoluminescence quantum yield of $\text{Ce}^{3+}$ -doped aluminium-silicate glasses for scintillation application

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# 1. Absorption and PL-PLE spectra



- The cut-off wavelength of the matrix glass is about 250 nm.
- The emission peak, excitation peak and the cut-off wavelength show a red-shift with the increase of Ce<sup>3+</sup> concentration, because the energy absorbed by the electronic transition gradually becomes weaker.
- PL decay times decrease with further increase of Ce<sup>3+</sup> ions concentration, which are the typical values for the 5d→4f transition of Ce<sup>3+</sup> ions.

## 2. PL QY

Glass composition	Density(g/cm <sup>3</sup> )	PL QY (%)
Ce <sup>3+</sup> -Li <sub>3</sub> PO <sub>4</sub> -B <sub>2</sub> O <sub>3</sub>	~	15.3
0.5Ce <sup>3+</sup> -55B <sub>2</sub> O <sub>3</sub> -20CaO-10Al <sub>2</sub> O <sub>3</sub> -15La <sub>2</sub> O <sub>3</sub>	~	42
0.1Ce <sup>3+</sup> -33.3Li <sub>2</sub> O-66.7SiO <sub>2</sub>	~	33
0.7Ce <sup>3+</sup> -34Li <sub>2</sub> O-5MgO-10Al <sub>2</sub> O <sub>3</sub> -51SiO <sub>2</sub> (Commercial GS20)	2.5	~90
0.1Ce <sup>3+</sup> -40BaO-60B <sub>2</sub> O <sub>3</sub>	~3.75	40
YAG: Ce <sup>3+</sup> (as a standard sample)	~	87.22
0.3Ce <sup>3+</sup> -doped 20Gd <sub>2</sub> O <sub>3</sub> -20Al <sub>2</sub> O <sub>3</sub> -60SiO <sub>2</sub>	~4.2	28.32
0.7Ce <sup>3+</sup> -doped 20Gd <sub>2</sub> O <sub>3</sub> -20Al <sub>2</sub> O <sub>3</sub> -60SiO <sub>2</sub>	~4.2	38.90
1.1Ce <sup>3+</sup> -doped 20Gd <sub>2</sub> O <sub>3</sub> -20Al <sub>2</sub> O <sub>3</sub> -60SiO <sub>2</sub>	~4.2	50.50
1.5Ce <sup>3+</sup> -doped 20Gd <sub>2</sub> O <sub>3</sub> -20Al <sub>2</sub> O <sub>3</sub> -60SiO <sub>2</sub>	~4.2	48.98
1.9Ce <sup>3+</sup> -doped 20Gd <sub>2</sub> O <sub>3</sub> -20Al <sub>2</sub> O <sub>3</sub> -60SiO <sub>2</sub>	~4.2	38.32

$$LY = \frac{E}{\beta E_g} * S * Q$$

LY—light yield of a scintillator

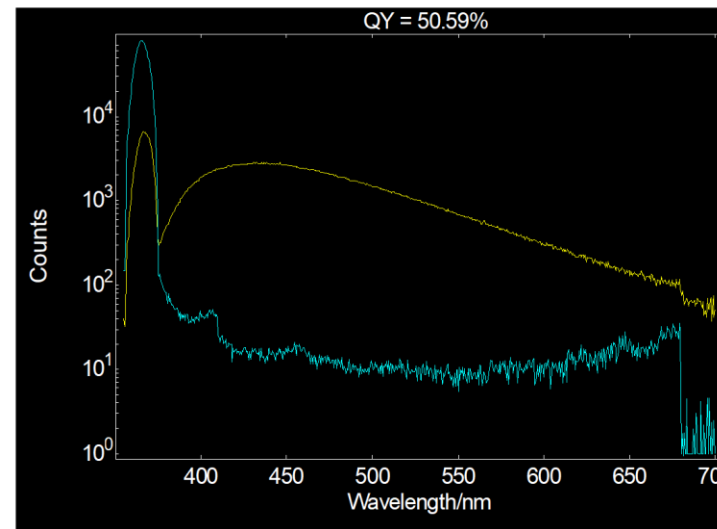
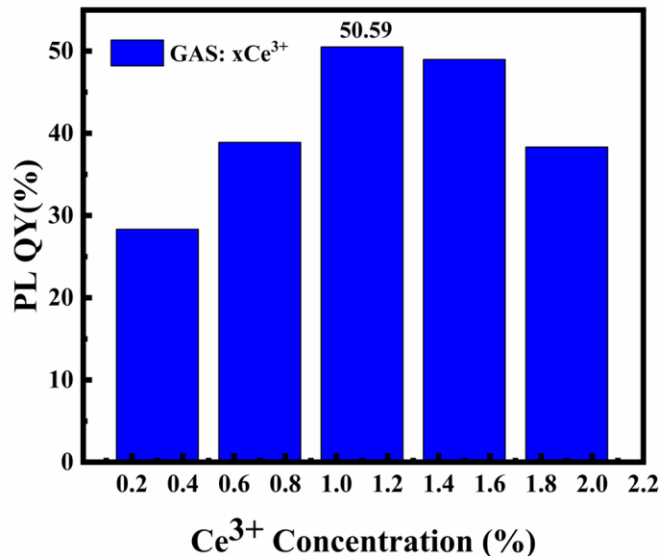
E—deposited energy of ionizing radiation

$\beta$ —constant parameter dependent on host material

$E_g$ —band gap energy

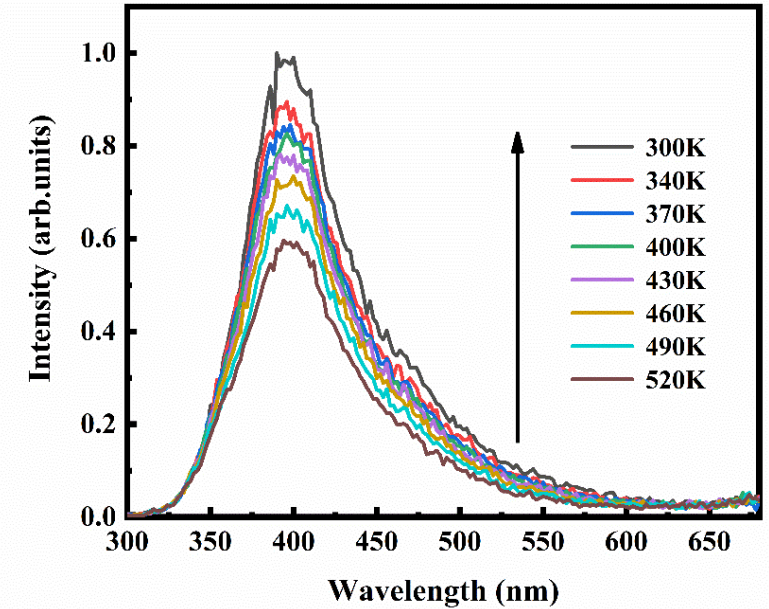
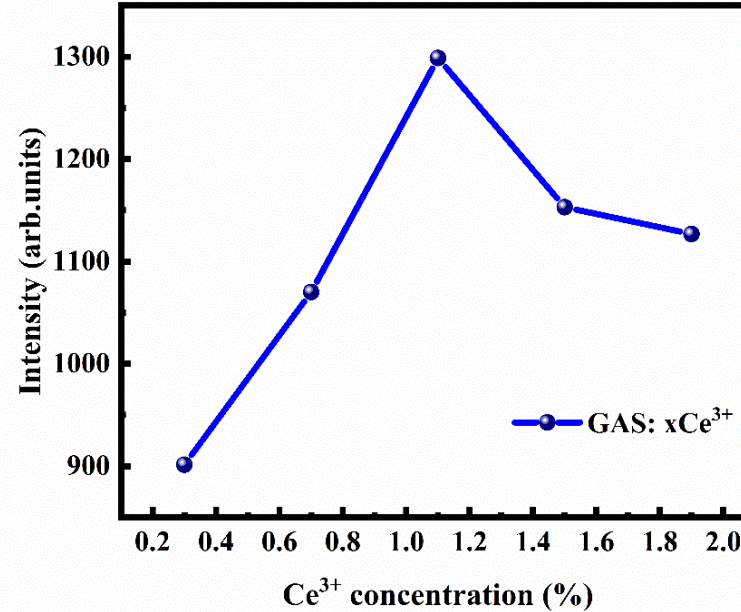
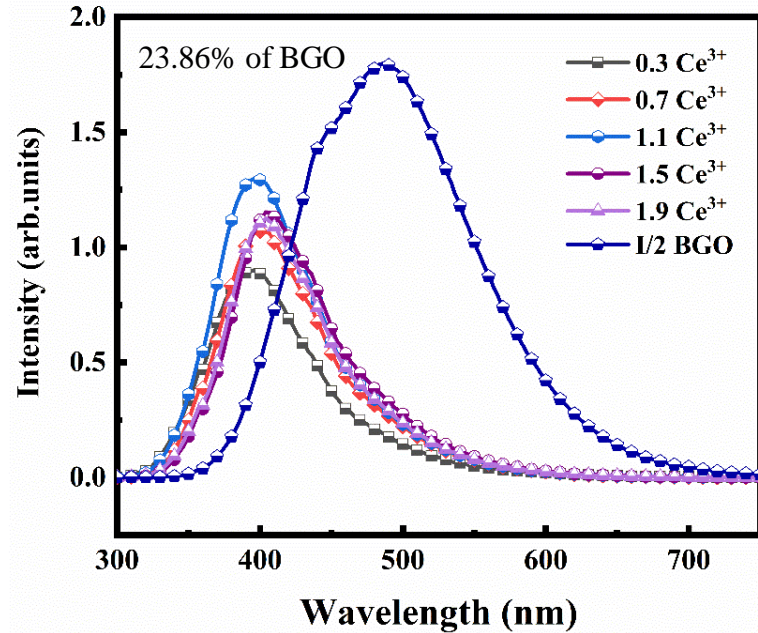
S—energy migration

Q—photoluminescence quantum yield



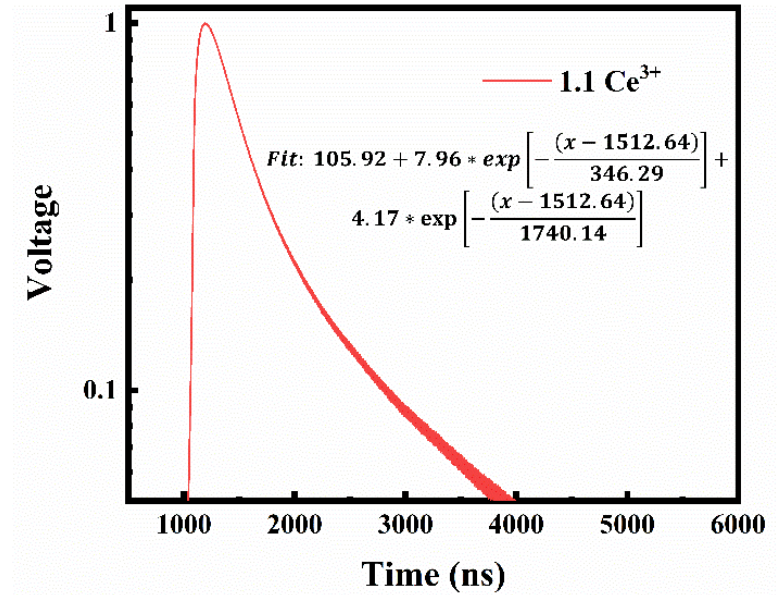
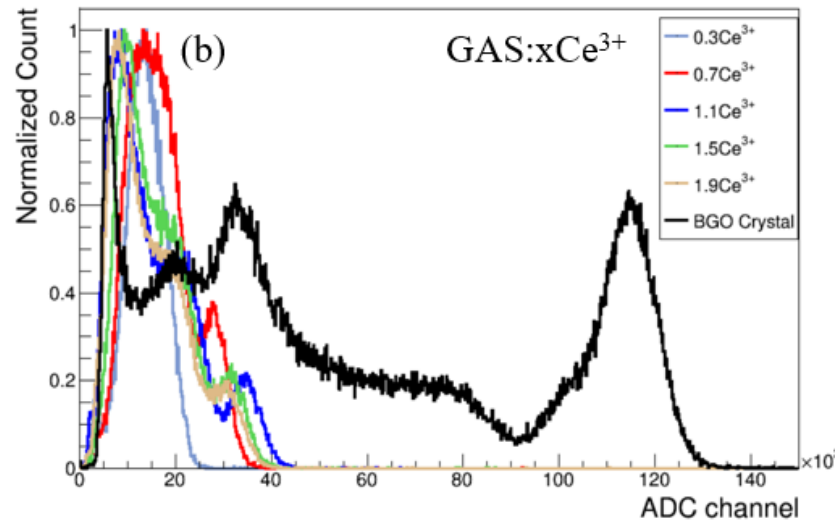
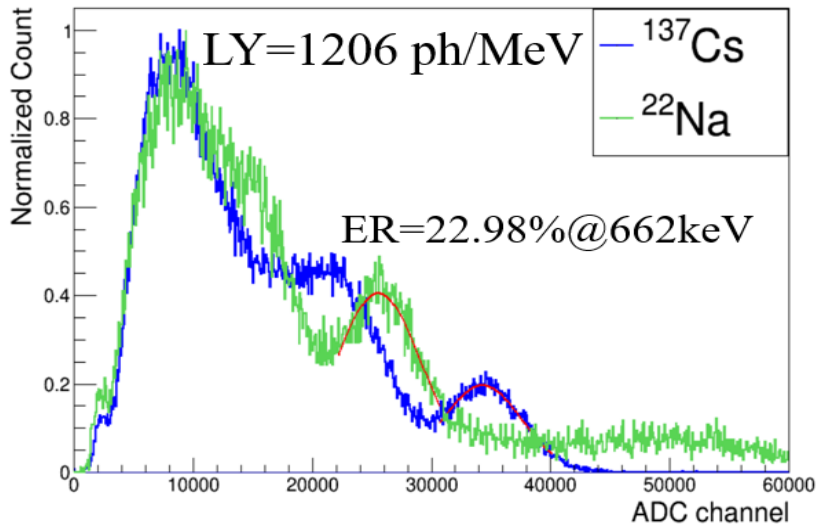
- The PL QYs of GAS: xCe<sup>3+</sup> glasses first increase and then decrease with the increase of Ce<sup>3+</sup> concentration, and reach the maximum when x=1.1 mol%.
- When the distance among Ce<sup>3+</sup> ions exceed a certain threshold, the interaction becomes dominant, resulting in concentration quenching.

### 3. XEL and thermal quenching



- The emission peak positions of the glasses are different in XEL (390-400 nm) and PL (420-440 nm) spectra due to the difference of the luminescent mechanism and excitation energy.
- The GAS:1.1Ce<sup>3+</sup> glass shows the highest intensity which is approximately 23.86% that of BGO crystal.
- At 520 K, the luminescent intensity of GAS:1.1Ce<sup>3+</sup> glass is 59.67% of that of room temperature (300 K), and the thermal quenching in XEL is estimated to be all about 0.14 eV.

# 4. Gamma-ray scintillation



- The light yield of the GAS: 1.1Ce<sup>3+</sup> glass is calculated to be about 1206 ph/MeV with an energy resolution of 22.98% at 662keV.
- Under the same measurement conditions, the number of photons of glasses detected by SiPM is about 1/4-1/3 of BGO crystal. This result is consistent with change in XEL integral intensity.
- The decay time of the glasses consists of fast and slow components, which are between 395-285 ns and 2332-1382 ns. The fast component originates from the direct capture of electrons by Ce<sup>3+</sup> ions and the slow component originates from the repeated capture of electrons by defect levels.

Thank you!