

Study of the wall painting from the Vladychinaya palata of the Novgorod kremlin (Velikiy Novgorod, Russia) using complementary physico-chemical methods

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Outline

- Introduction and objectives
- Study of plasters
- Study pigment composition in paint layers
- Conclusions

Vladychinaya palata

1. The UNESCO World Heritage Site;
2. It was built in 1433 with the participation of German masters, in the Western European Gothic style;
3. It served as the meeting place of the council of nobility and the boyar court of the Novgorod republic;
4. Interior decorated with 15th-19th century wall painting.





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Wall painting fragments

FLnP FRANK LABORATORY
OF NEUTRON PHYSICS



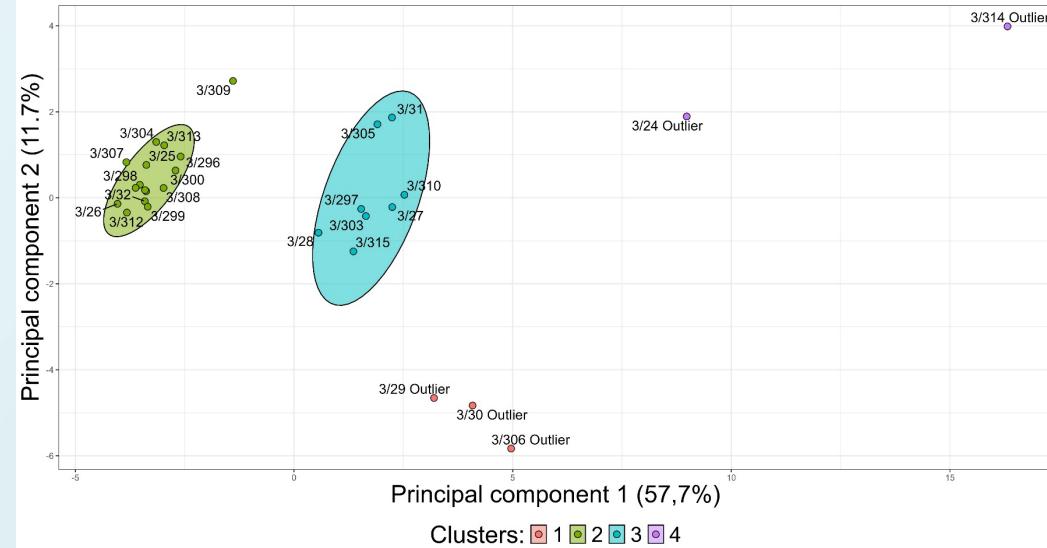
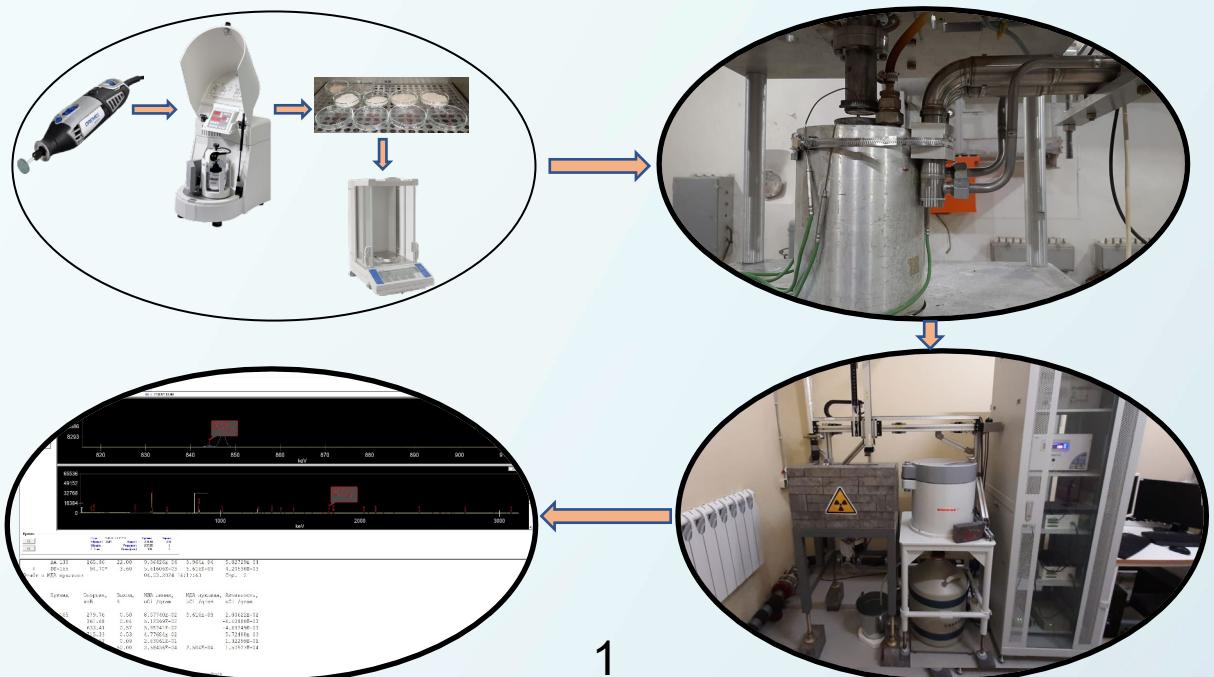
Comprehensive study of the wall painting
fragments in the Vladychinaya palata of the
Novgorod kremlin using complementary
physico-chemical methods

Tasks

1. Study the elemental composition of plasters;
2. Check the possibility of grouping samples by elemental composition using multivariate mathematical statistics methods;
3. Study the pigment composition of paint layers;
4. Carry out a comprehensive description of the samples based on elemental and pigment analysis.

Study methods

1. Neutron activation analysis (NAA);
2. K-means and PCA methods;
3. X-ray fluorescence analysis (XRF);
4. Stratigraphic analysis;
5. Optical and polarized microscopy;
6. Raman spectroscopy.



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WWR-K reactor “Dry” channel for short-lived isotopes:

Neutron flux:

- thermal: $4.4 * 10^{12}$ n / cm²s;
- resonance: $3.8 * 10^{10}$ n / cm²s.

Weight of samples:

- 0.1 grams

Irradiation time:

- 1 minute

WWR-K reactor “Wet” channel for long-lived isotopes:

Neutron flux:

- thermal: $6.6 * 10^{13}$ n / cm²s;
- resonance: $3.0 * 10^{12}$ n / cm²s.

Weight of samples:

- 0.1 grams

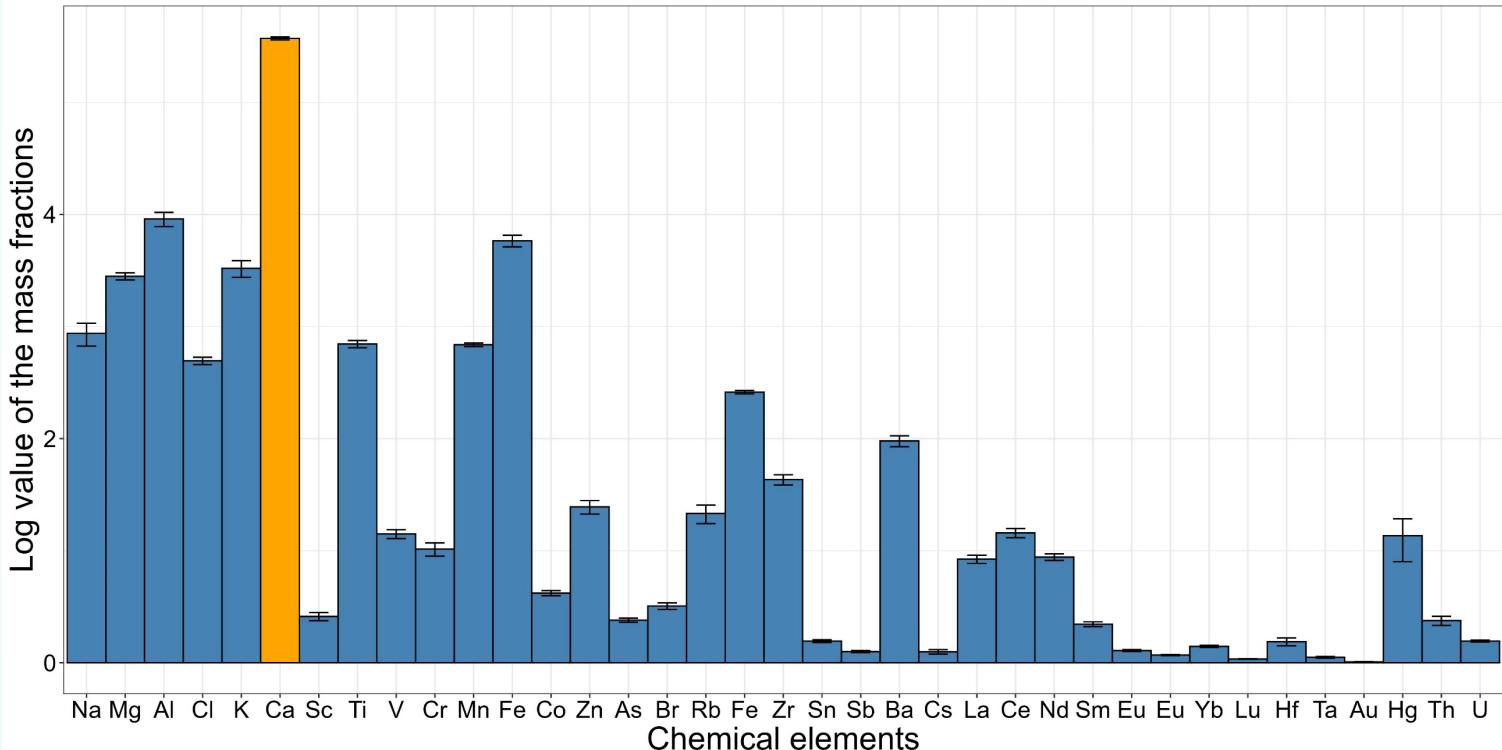
Irradiation time:

- 90 minutes

Neutron Activation Analysis

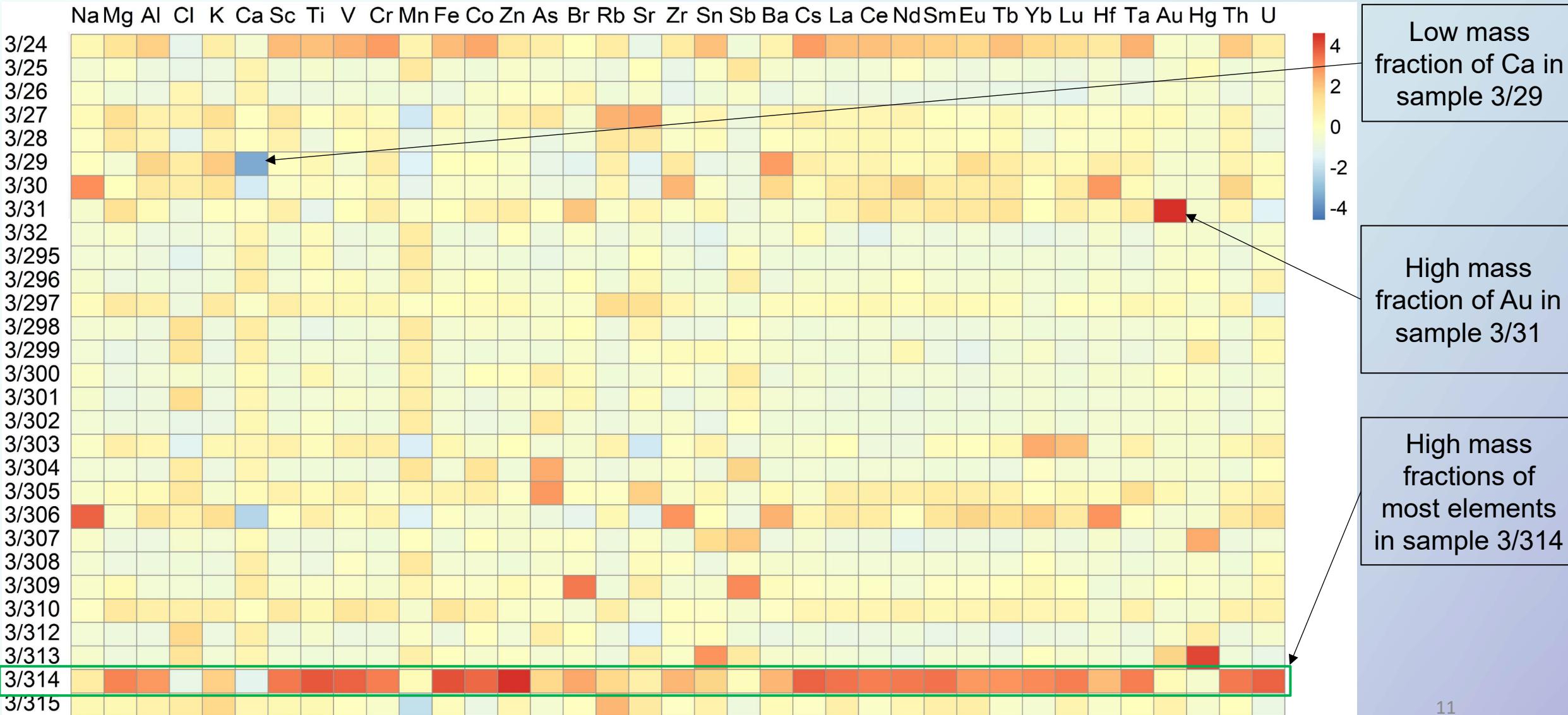
1. Relative NAA method;
2. NIST Standards: 1486, 1566B, 1632E, 1633C, 2586, 2706, 2709A, 2710A, 2780A;
3. Mass fractions of 37 chemical elements.

Mass fraction distribution of chemical elements



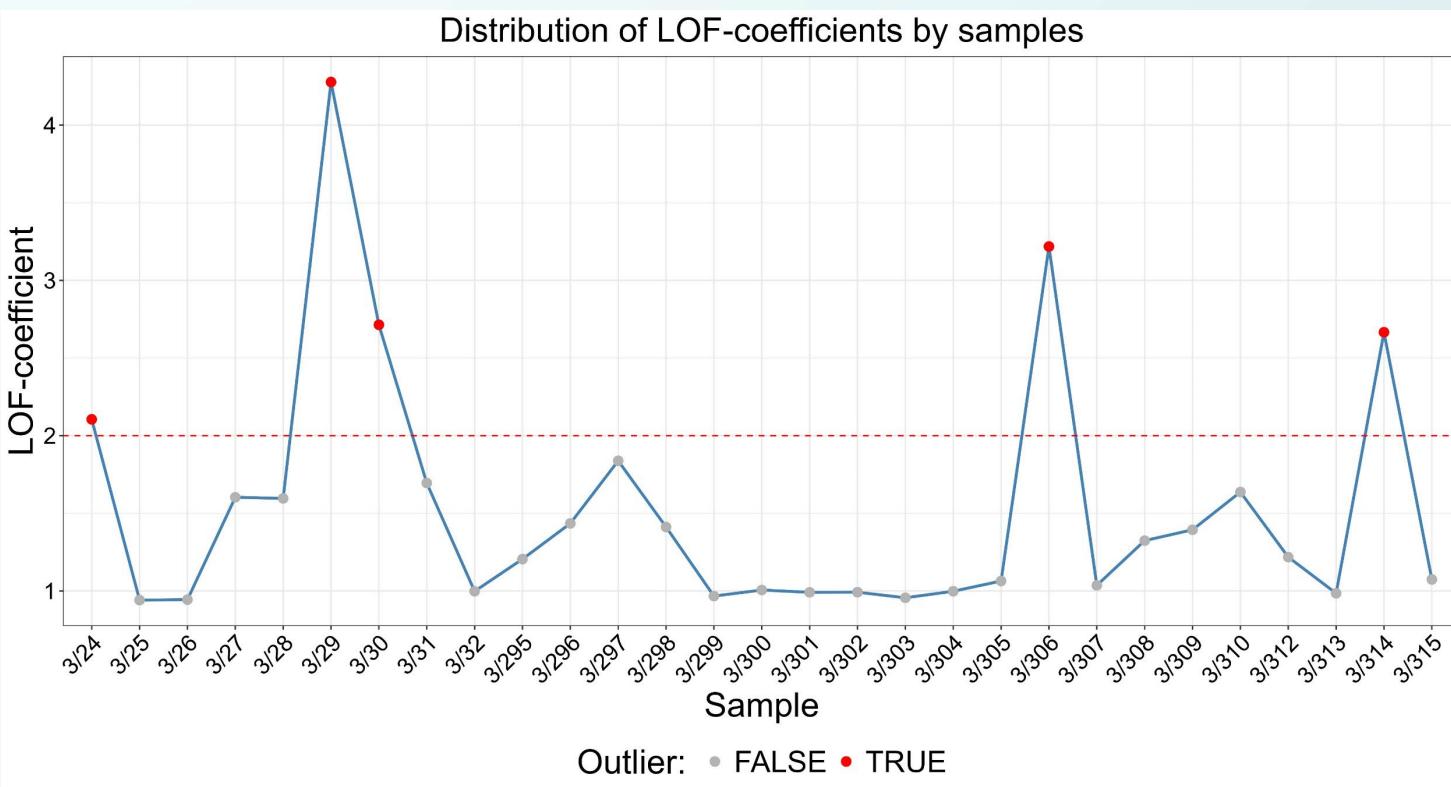
Element	Mean±Se	Median	Min–Max
Na	869±200	479	207–4870
Mg	2810±203	2430	1640–6250
Al	9130±1330	4580	2610–29000
Cl	494±37.5	410	217–820
K	3310±562	1480	442–9120
Ca	373000±11700	397000	157000–429000
Sc	1.58±0.216	0.915	0.551–5.47
Ti	698±52.1	656	384–1800
V	13.1±1.27	11.5	7.20–38.6
Cr	9.33±1.39	5.52	3.70–34.04
Mn	687±26.4	750	398–865
Fe	5830±692	4700	2740–20900
Co	3.17±0.218	2.79	2.08–7.40
Zn	23.6±3.39	19.4	10.3–108
As	1.39±0.102	1.20	0.825–2.94
Br	2.20±0.219	1.94	0.714–6.15
Rb	20.5±4.05	5.45	2.67–71.5
Sr	259±9.18	262	172–383
Zr	42.1±4.50	37.0	15.2–113
Sn	0.552±0.0458	0.499	0.253–1.27
Sb	0.249±0.0290	0.196	0.108–0.724
Ba	94.3±10.6	70.8	40.7–250
Cs	0.243±0.0587	0.109	0.0317–1.42
La	7.40±0.714	5.35	3.75–20.6
Ce	13.4±1.36	9.37	4.32–37.6
Nd	7.76±0.593	7.45	3.80–18.4
Sm	1.20±0.109	0.904	0.59–3.23
Eu	0.277±0.0255	0.209	0.115–0.669
Tb	0.163±0.0137	0.130	0.0805–0.375
Yb	0.392±0.0301	0.340	0.220–0.887
Lu	0.0711±0.00606	0.0566	0.0301–0.179
Hf	0.532±0.126	0.223	0.0664–2.52
Ta	0.110±0.0179	0.0660	0.0254–0.427
Au	0.00883±0.00225	0.0049	0.00239–0.0645
Hg	12.6±5.64	1.14	0.134–140
Th	1.37±0.222	0.667	0.377–5.33
U	0.552±0.0289	0.514	0.330–1.14

Heatmap of element mass fractions

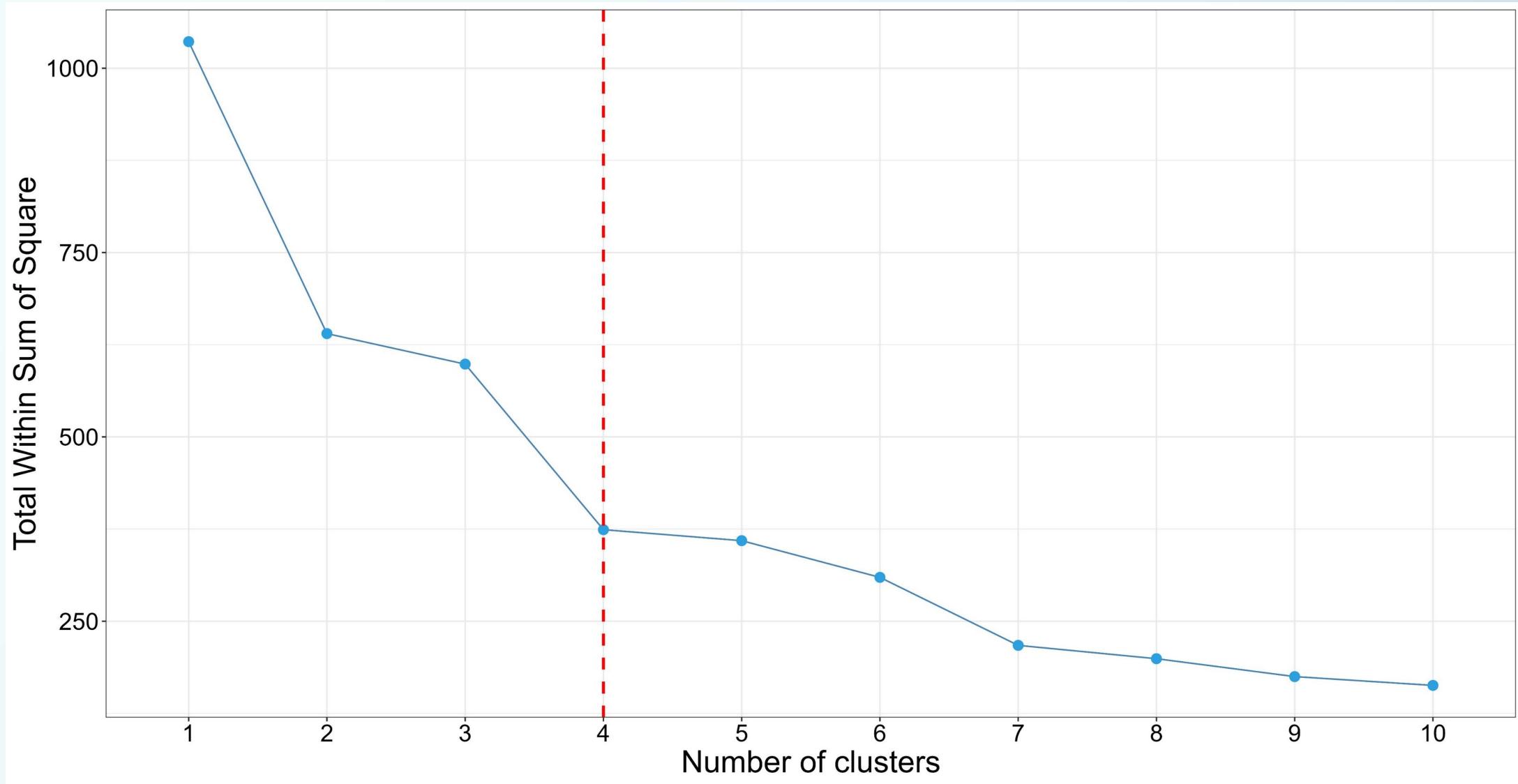


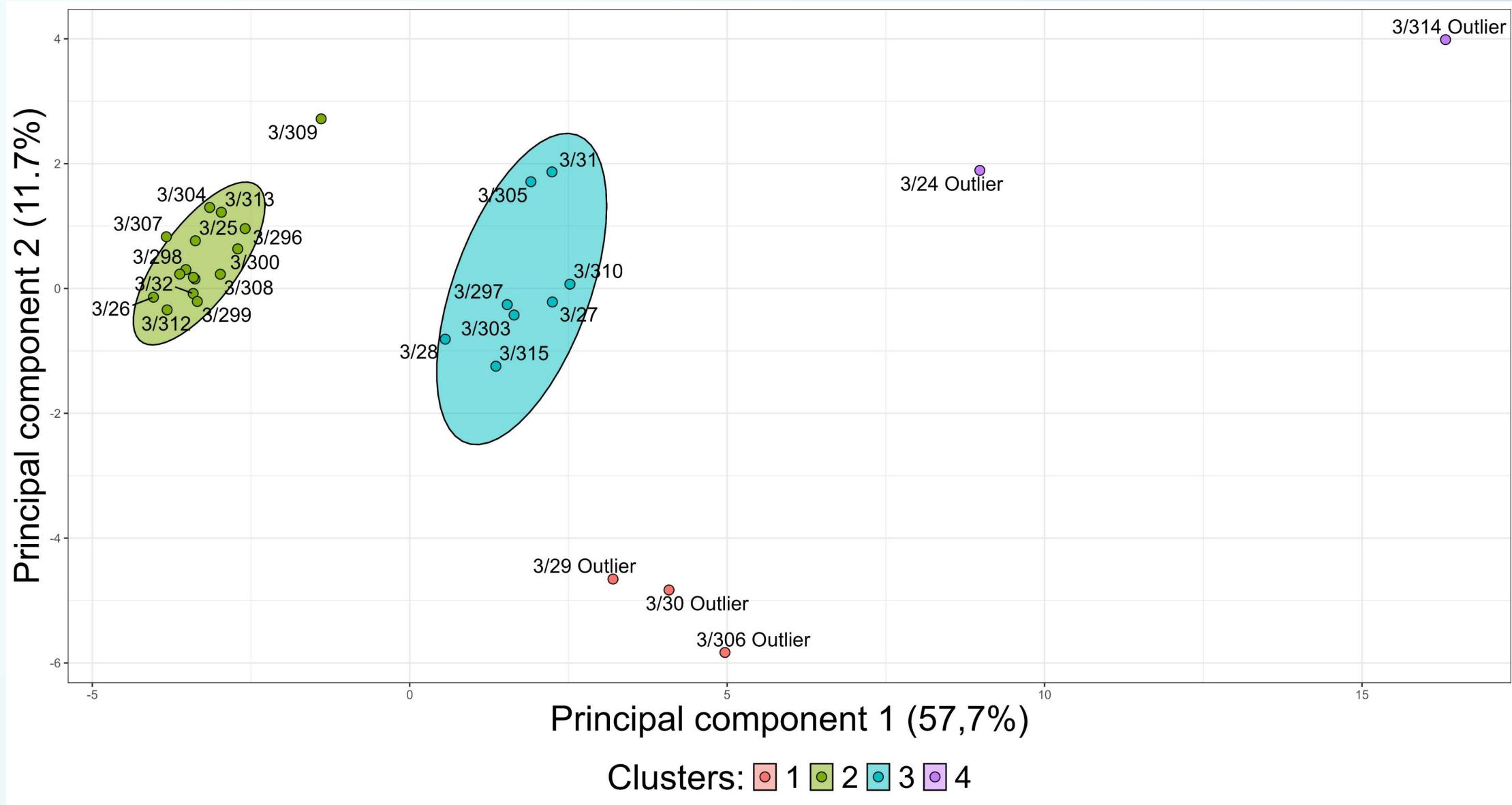
Outlier detection

1. Shapiro-Wilk and Henze-Zirkler tests: data aren't normally distributed;
2. Local Outlier Factor (LOF) method: outlier detection without distribution assumptions;
3. LOF-coefficient > 2 - optimal balance for outlier detection;
4. Detected outliers: 3/24, 3/29, 3/30, 3/306, 3/314.



Sample	LOF-coefficient	Outlier
3/24	2.105	TRUE
3/25	0.941	FALSE
3/26	0.944	FALSE
3/27	1.603	FALSE
3/28	1.596	FALSE
3/29	4.277	TRUE
3/30	2.714	TRUE
3/31	1.695	FALSE
3/32	0.998	FALSE
3/295	1.205	FALSE
3/296	1.435	FALSE
3/297	1.838	FALSE
3/298	1.412	FALSE
3/299	0.967	FALSE
3/300	1.006	FALSE
3/301	0.991	FALSE
3/302	0.992	FALSE
3/303	0.956	FALSE
3/304	0.998	FALSE
3/305	1.064	FALSE
3/306	3.218	TRUE
3/307	1.036	FALSE
3/308	1.324	FALSE
3/309	1.394	FALSE
3/310	1.637	FALSE
3/312	1.218	FALSE
3/313	0.985	FALSE
3/314	2.666	TRUE
3/315	1.074	FALSE

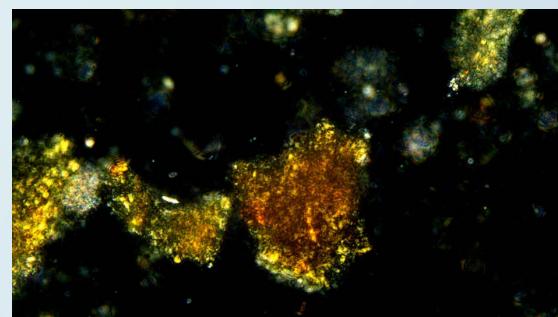
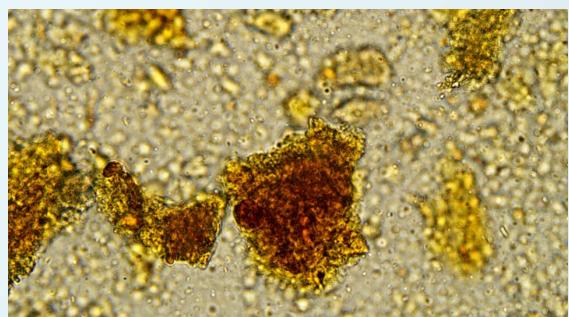
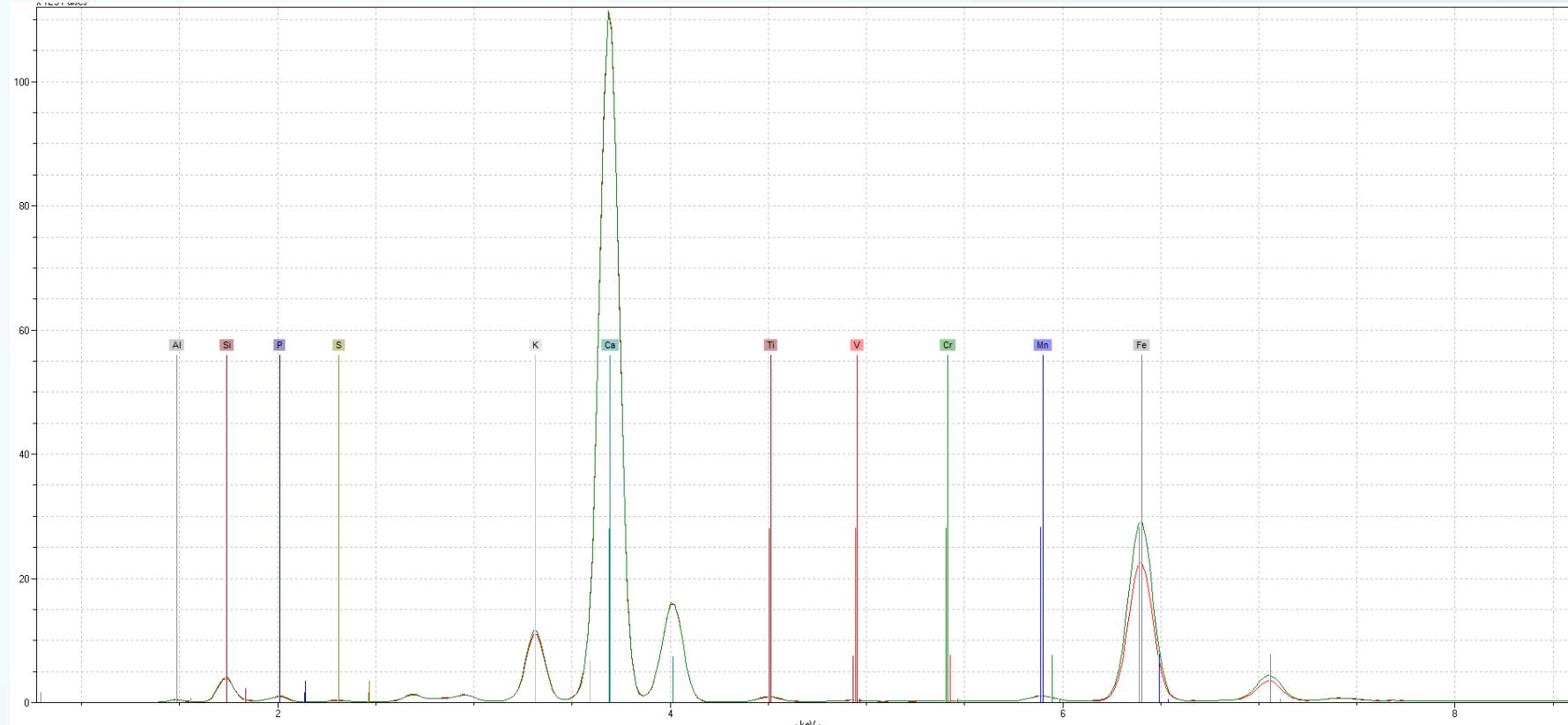




Outline

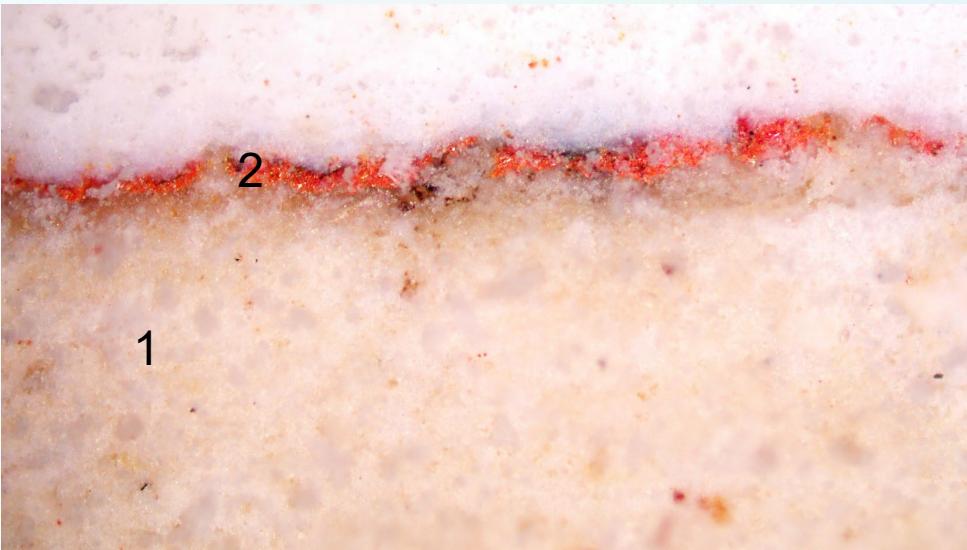
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Methods for studying pigment composition





Red paint layer

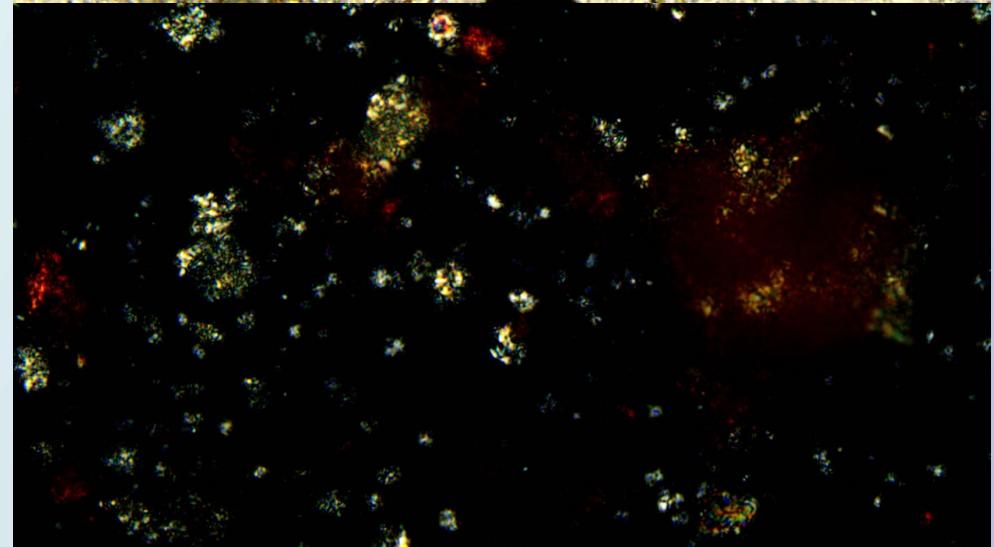
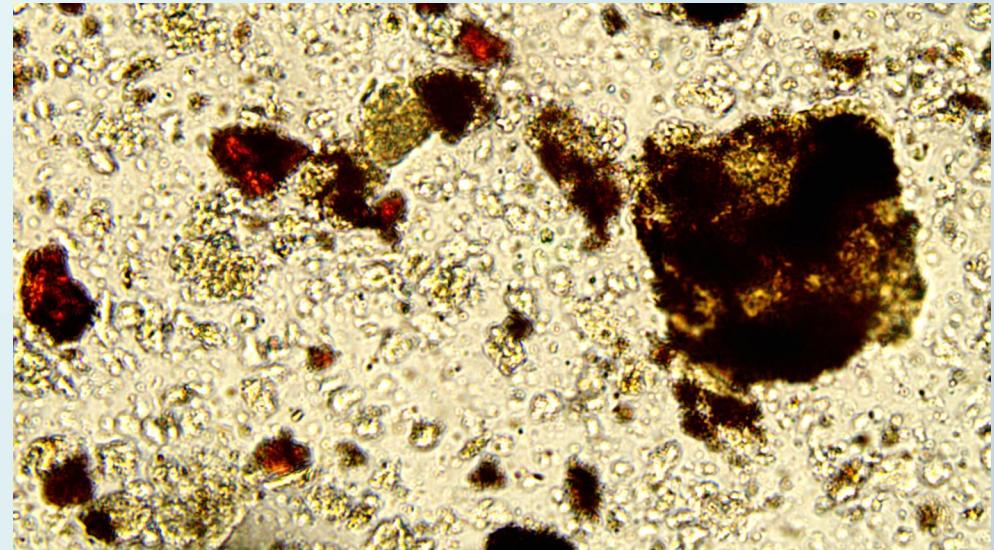


2. Red paint layer with black inclusions
1. Plaster

Color	Characteristic elements
Red	Al, Si, Hg , K, Ca, Ti, Mn, Fe

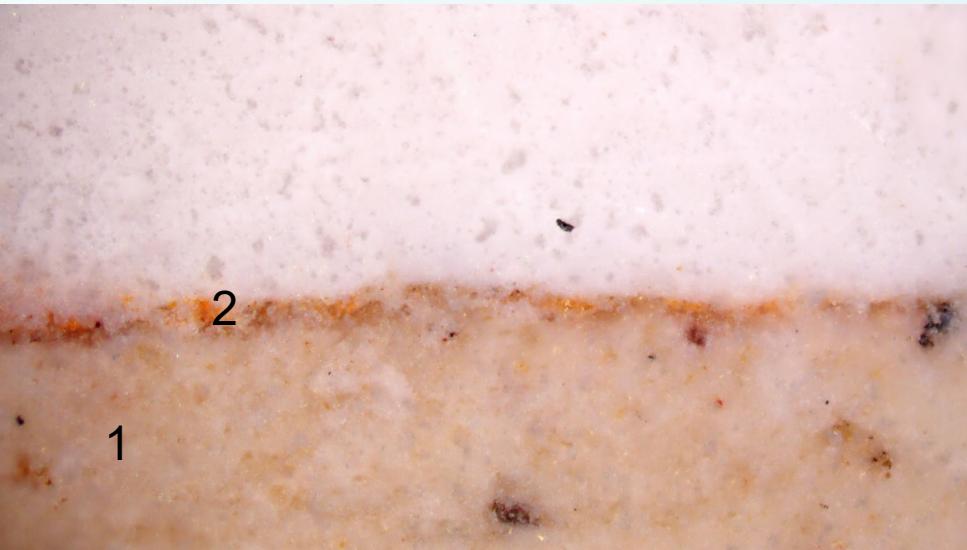
Pigments:

- Red ochre (Fe_2O_3)
- Cinnabar (HgS)
- Carbon black (C)





Orange paint layer

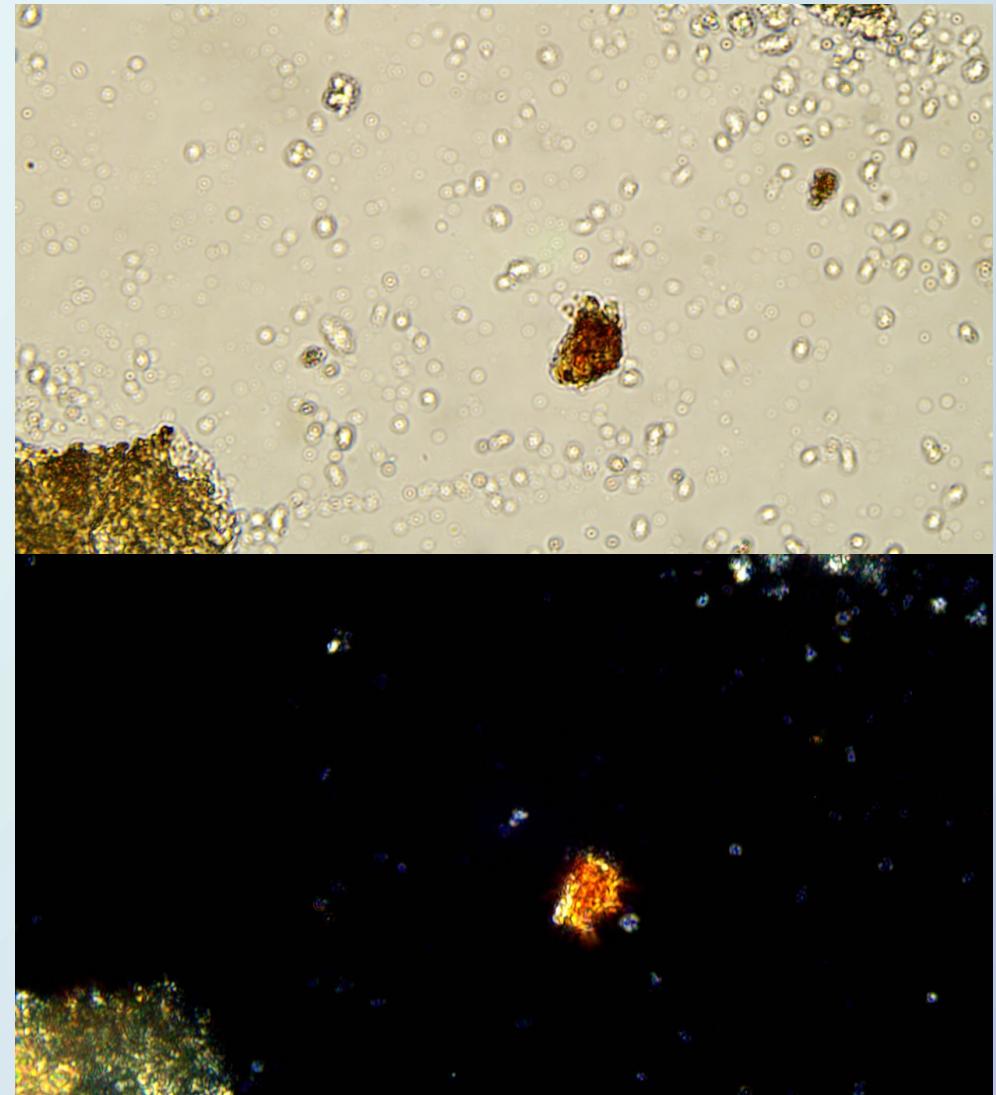


2. Orange paint layer
1. Plaster

Color	Characteristic elements
Orange	Al, Si, Pb , K, Ca, Ti, Mn, Fe

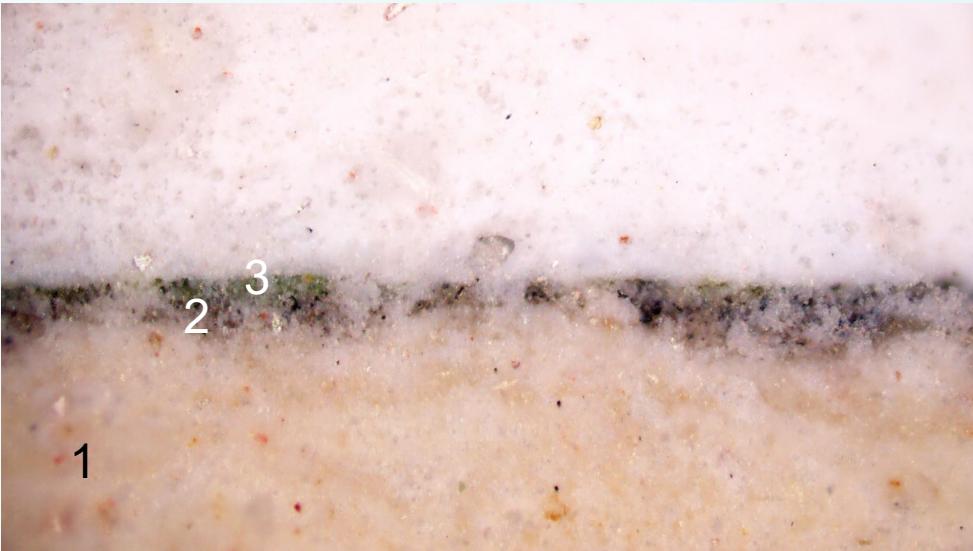
Pigments:

- Red lead (Pb_3O_4)





Green paint layer

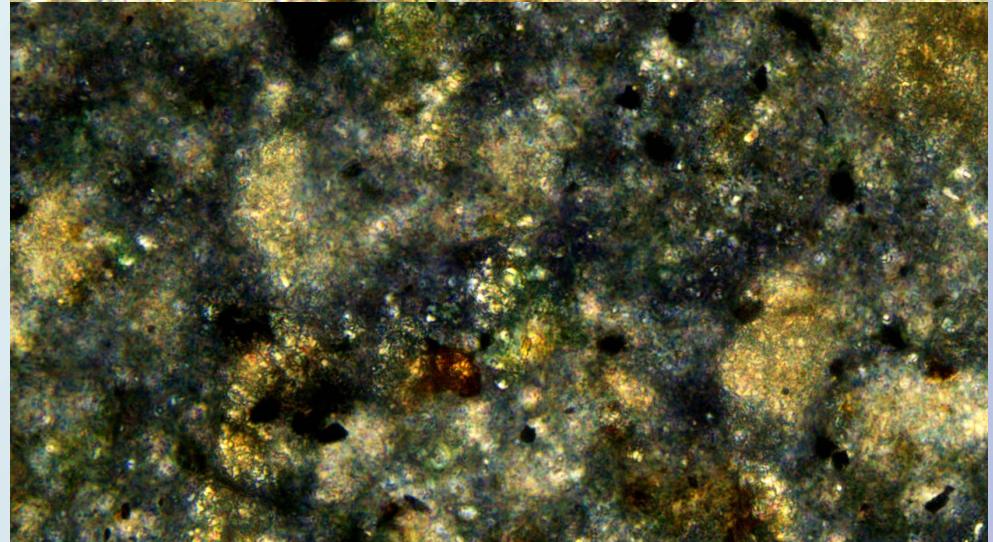
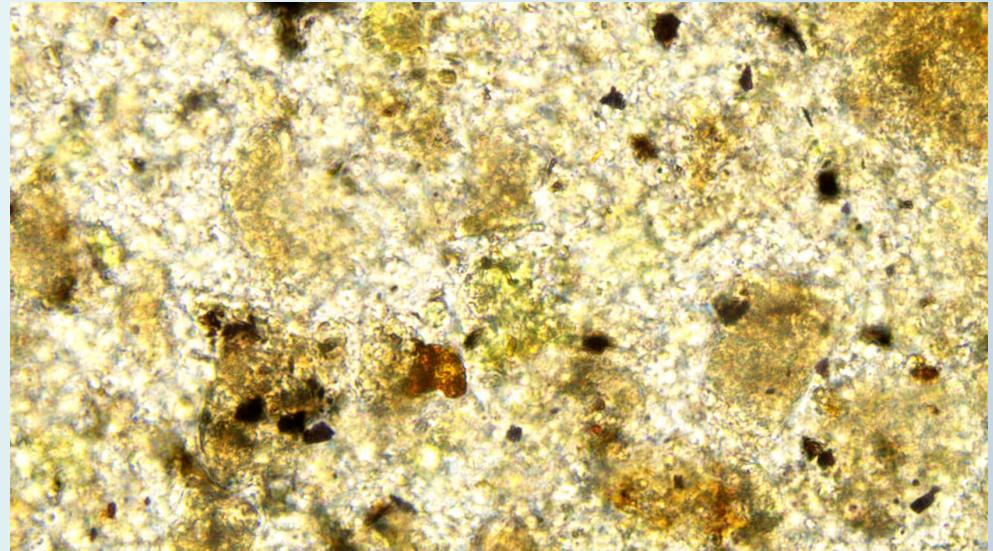


3. Green paint layer
2. Black paint later (Reft)
1. Plaster

Color	Characteristic elements
Dark green	Al, Si, S, K, Ca, Ti, Mn, Fe

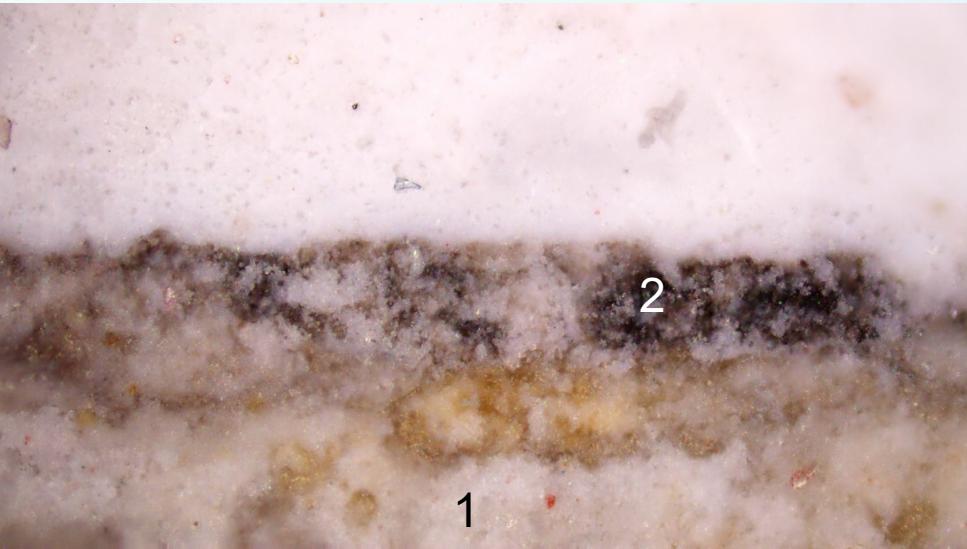
Pigments:

- Green earth
 $(\text{Si}_{4x}\text{Al}_x)_4(\text{Fe}^{3+}, \text{Fe}^{2+}, \text{Al}, \text{Mg})_2\text{O}_{10}(\text{OH})_2\text{K}_{(x+y)}$
- Carbon black (C)





Black paint layer with fragmentary preserved blue layer

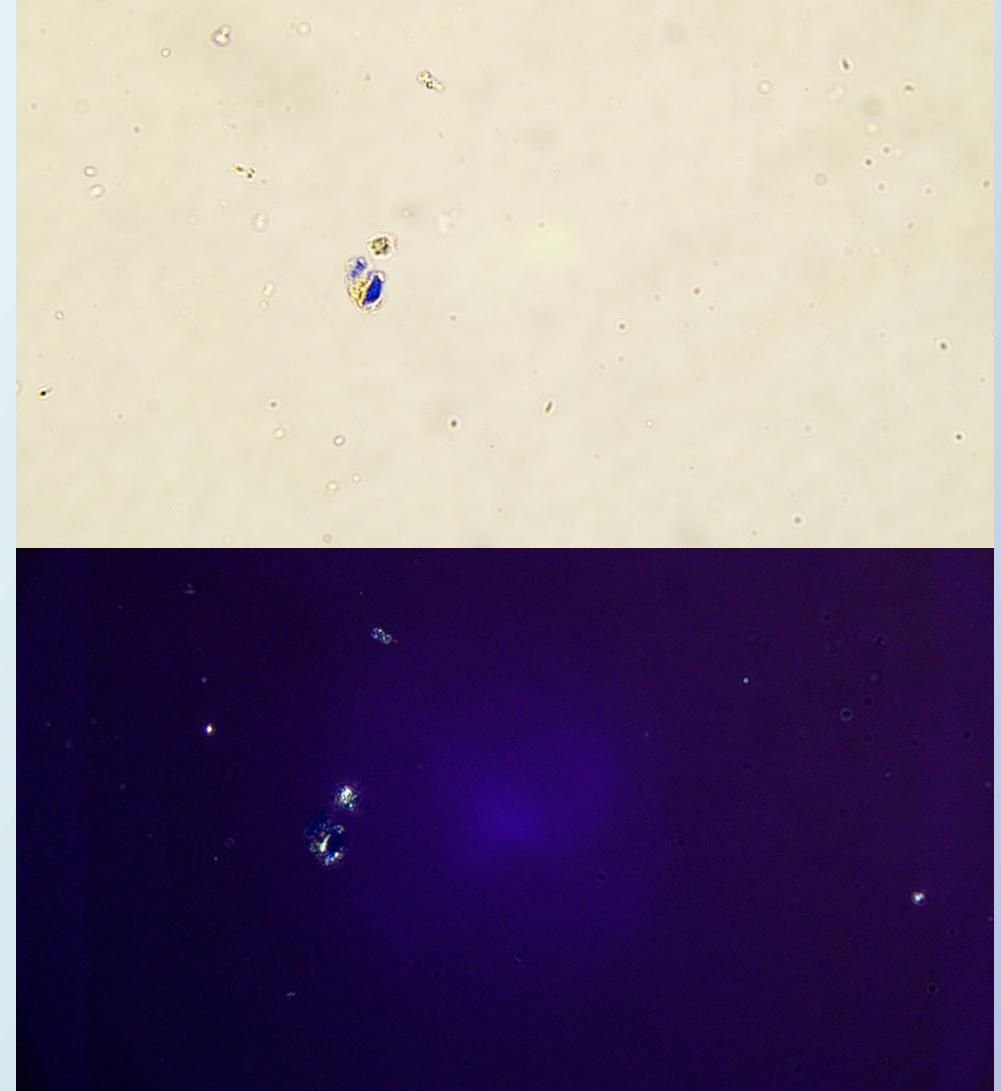


2. Grey paint layer
1. Plaster

Color	Characteristic elements
Grey	Al, Si, S, K, Ca, Ti, Mn, Fe

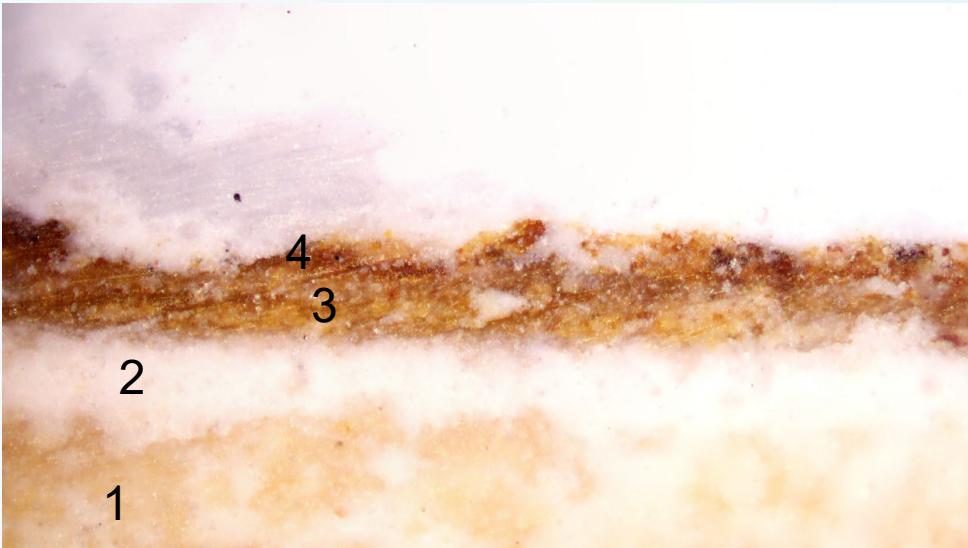
Pigment:

- Carbon black (C)
- Lime white (CaCO_3)
- Lazurite $(\text{Na,Ca})_8[(\text{Al},\text{Si})_{12}\text{O}_{24}](\text{S,SO}_4)$





Yellow paint layer

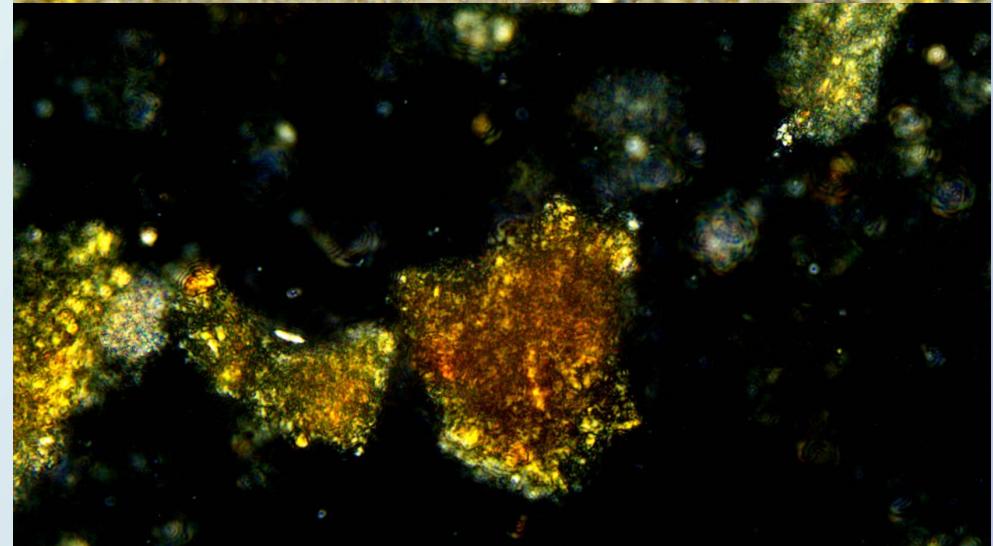
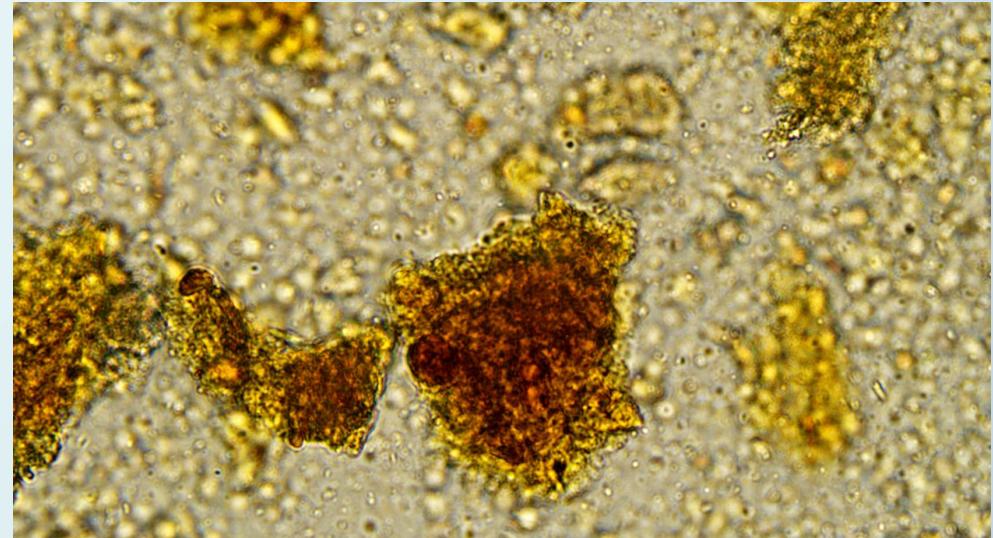


4. Dark yellow paint layer
3. Yellow paint layer
2. White paint layer
1. Plaster

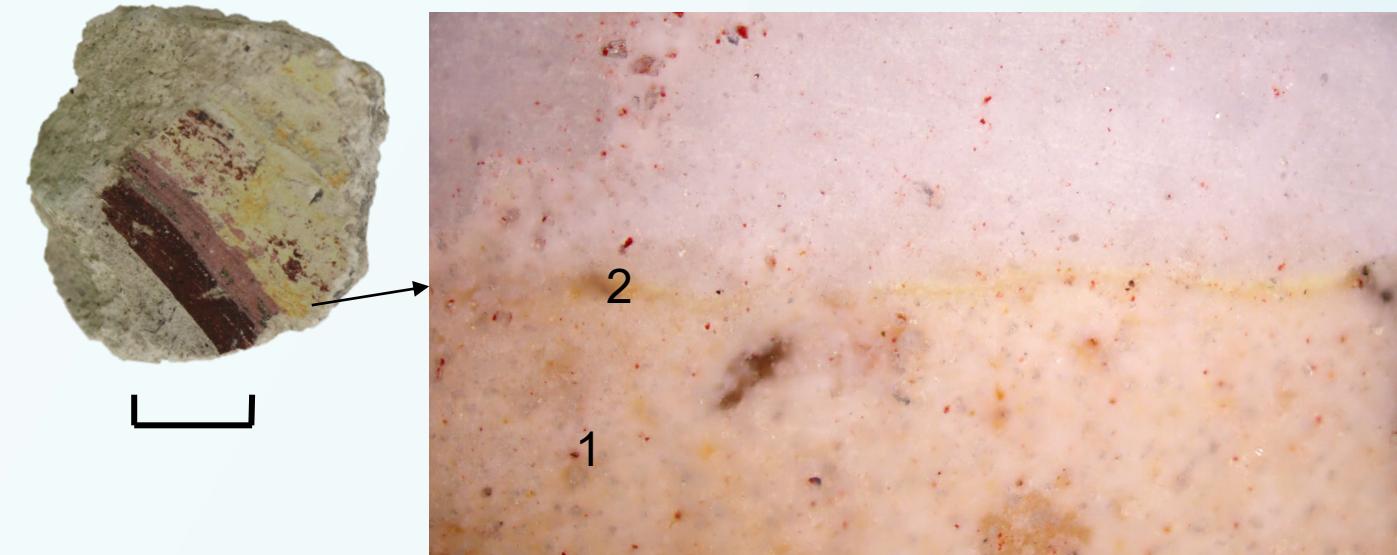
Color	Characteristic elements
Yellow	Al, Si, P, S, K, Ca, Ti, V, Mn, Fe

Pigment:

- Yellow ochre (FeO(OH))
- Lime white (CaCO_3)



Interesting case

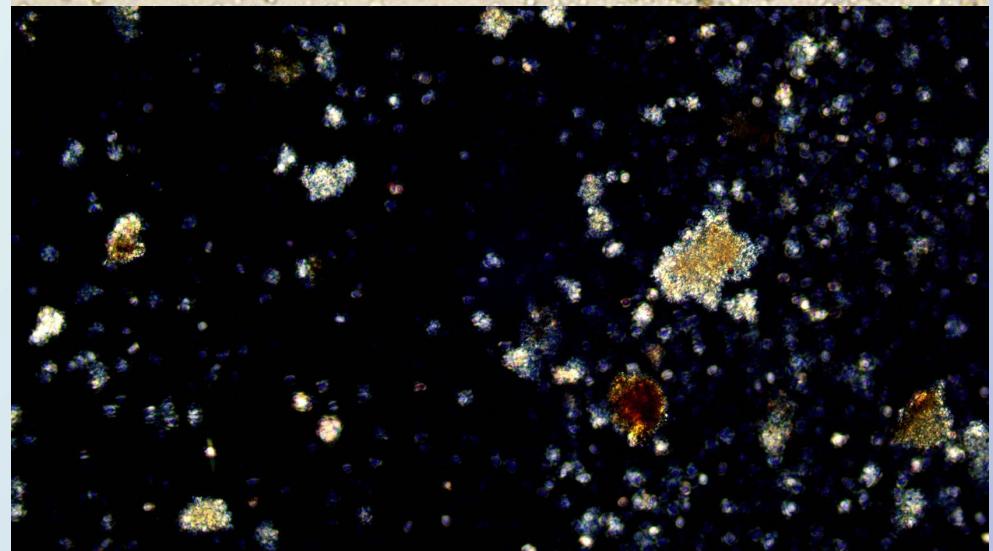
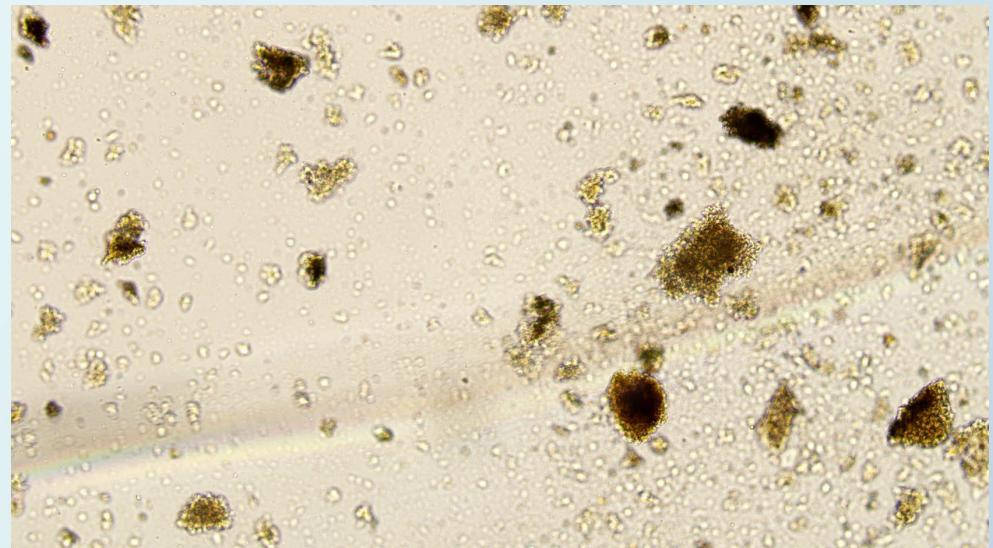


2. Yellow paint layer
1. Plaster

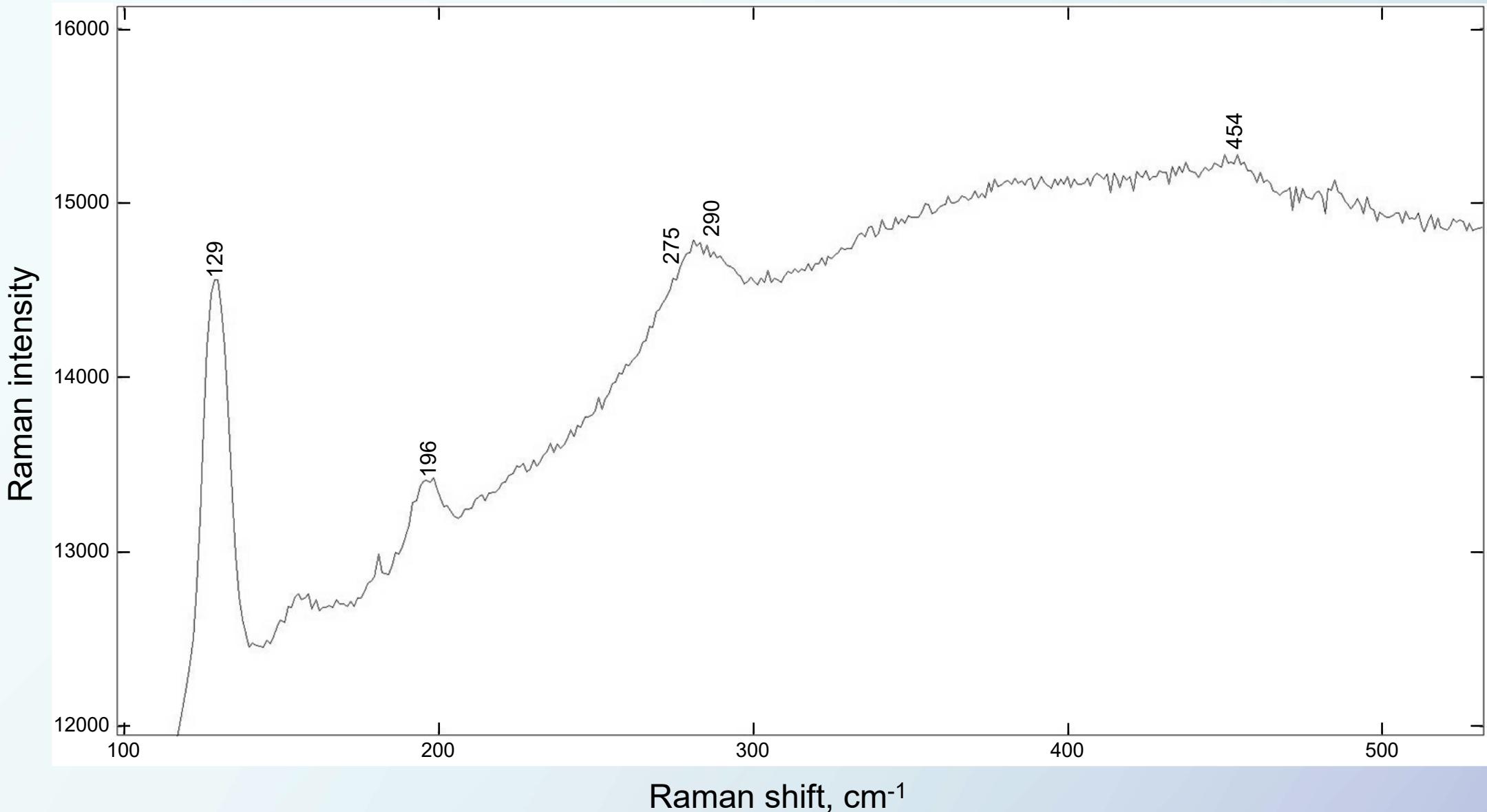
Color	Characteristic elements
Yellow	Al, Si, Pb , Sn, Ca, Mn, Fe

Pigment:

- Lead-tin yellow I (Pb_2SnO_4)



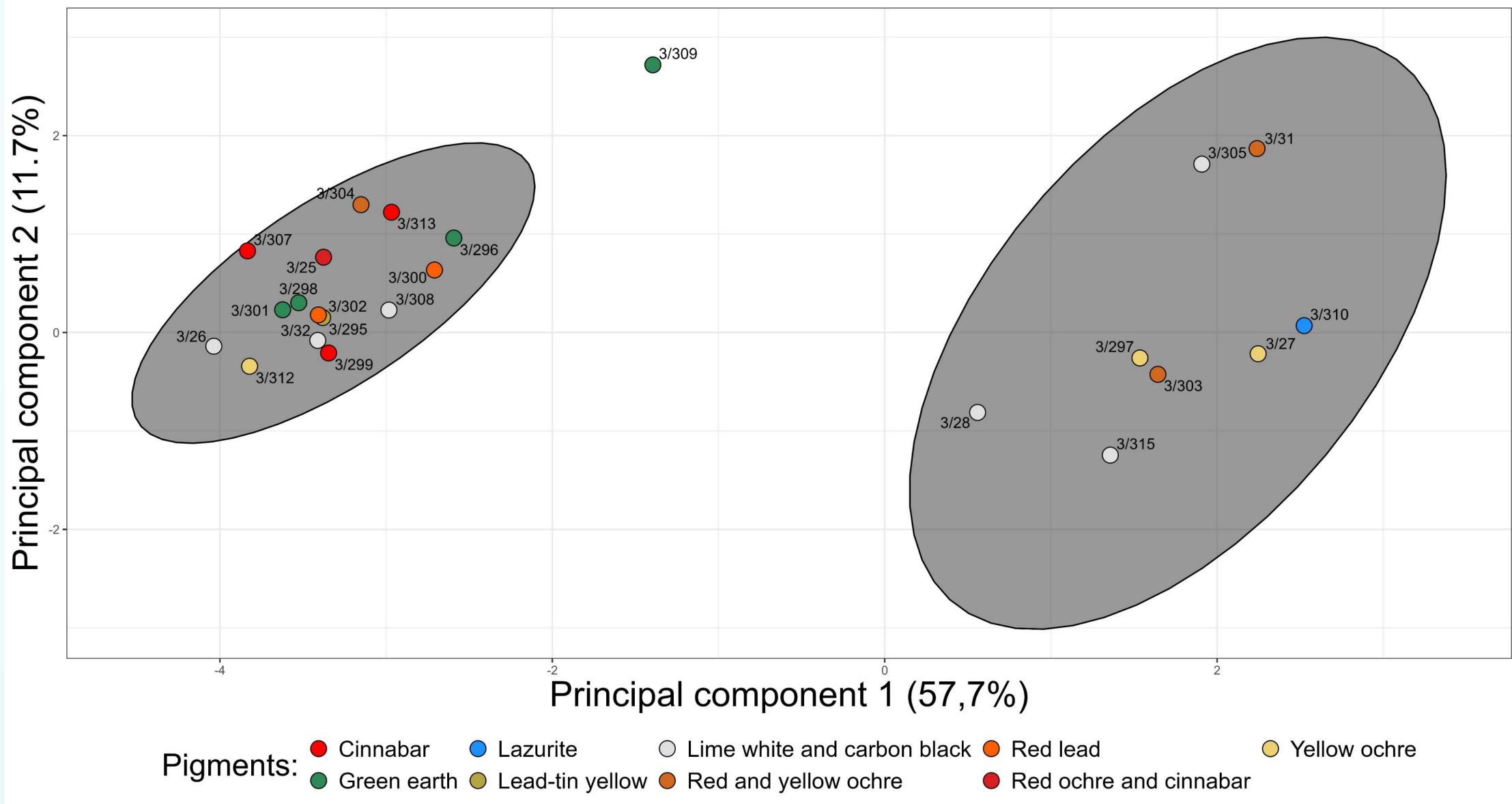
Lead-tin yellow I



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PCA method for cluster visualization with pigment color-coding





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Thank you for your attention!

Histogram of LOF-coefficients distribution

