

Research on Enhancing the Performance of Glass Scintillator



闪烁玻璃合作组
Glass Scintillator Collaboration



Sen Qian

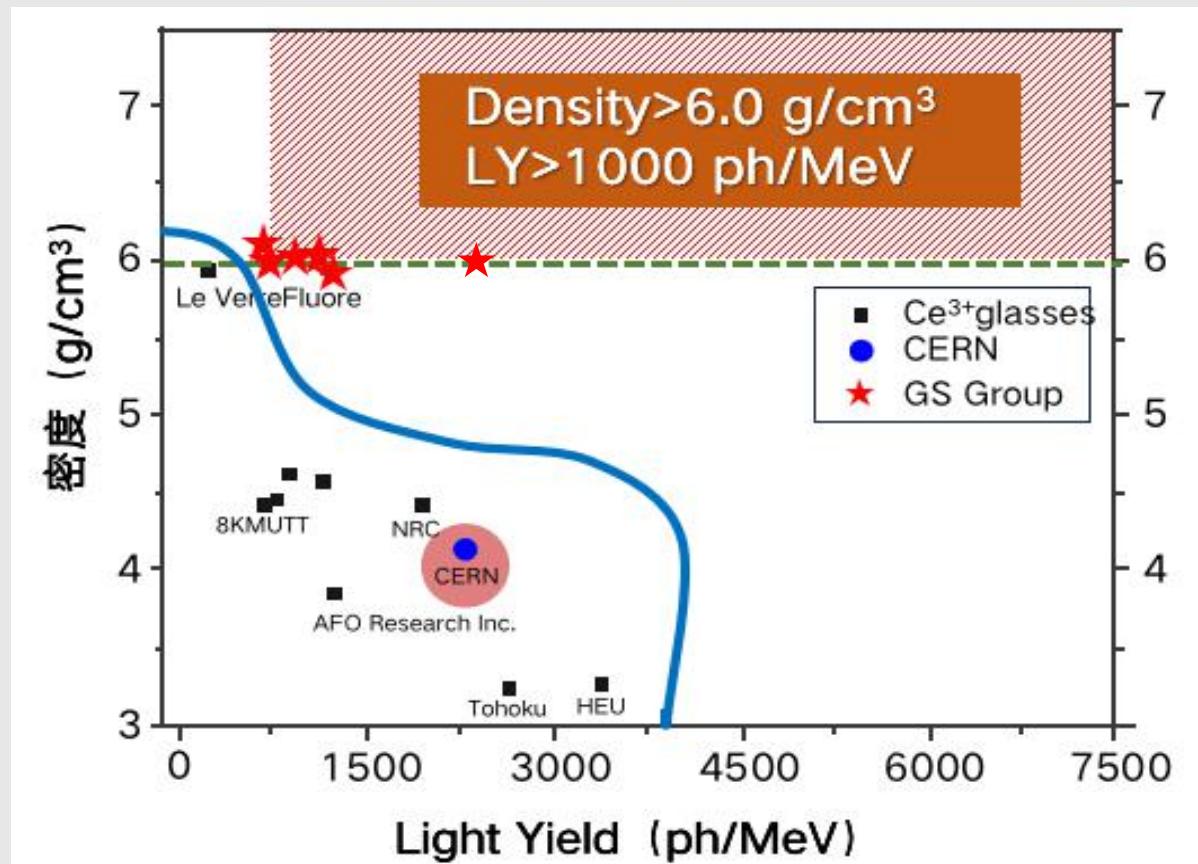
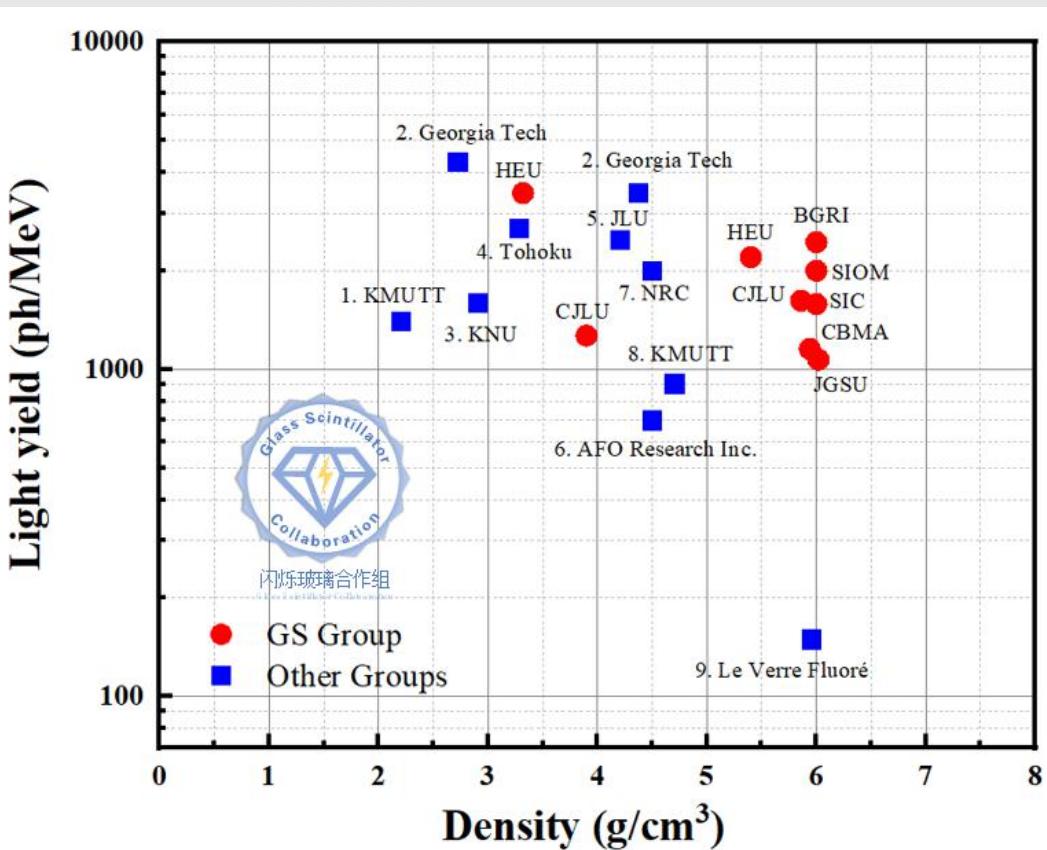
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The Institute of High Energy Physics, CAS

Outline

- 1. Performance target of Glass Scintillator (GFO)
- 2. Theoretical feasibility of Light Yield Enhancement
- 3. Challenges in Enhancing Light Yield
 - 3.1. Density
 - 3.2. Scintillation Decay Time
 - 3.3. Wavelength Shift
 - 3.4. Co-doping

1.0 Research Status of GS

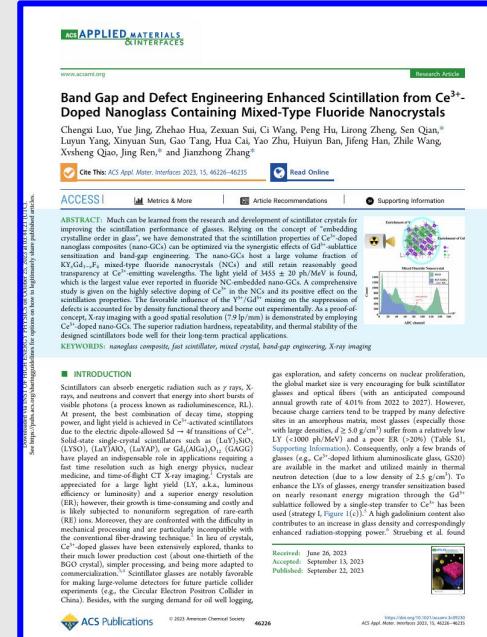


- The GS group has carried out a comprehensive and complete study;
- For high density glass scintillator, the light yield of GS group samples is in the absolute lead.

1.1 Performance target of GS

	2021.09 (1 st Meeting)	2024.05 (7 th Meeting)	2024.10 (8 th Meeting)	2025.09 (9 th Meeting)	2026
	Target	Realization	Target	Realization	Target
Dimension (mm ³)	/	5 × 5 × 5	40 × 40 × 10	40 × 40 × 10	40 × 40 × 10
Density (g/cm ³)	6.0	5.93	6.0	>6.0	6.0
Transmittance (% @400 nm)	>75	~70	>80	76	>80
Emission peak (nm)	~400	~400	~400	390	~400
Light yield (ph/MeV)	1000	985	1500	1408	2000
Energy resolution (% @662keV)	/	30.3	/	28.7	<25
Decay time (ns)	<100	35.8(10.7%), 105.1	<500	72.4 (10.5%), 547.8	<500
Attenuation length (cm@400nm)	/	/	/	6.05	>10

Published Paper about Glass Scintillator



➤ 2015 NIMA.
DSB:Ce glass
Density=4.2 g/cm³
LY=2500 ph/MeV

➤ 2020 NIMA.
AFO glass
Density=4.6 g/cm³
LY=700 ph/MeV

➤ 2022 J. Non-Cryst. Solids.
(Gd,Ce)₂O₃-Al₂O₃-SiO₂ glass
Density=4.5 g/cm³
LY=2000 ph/MeV

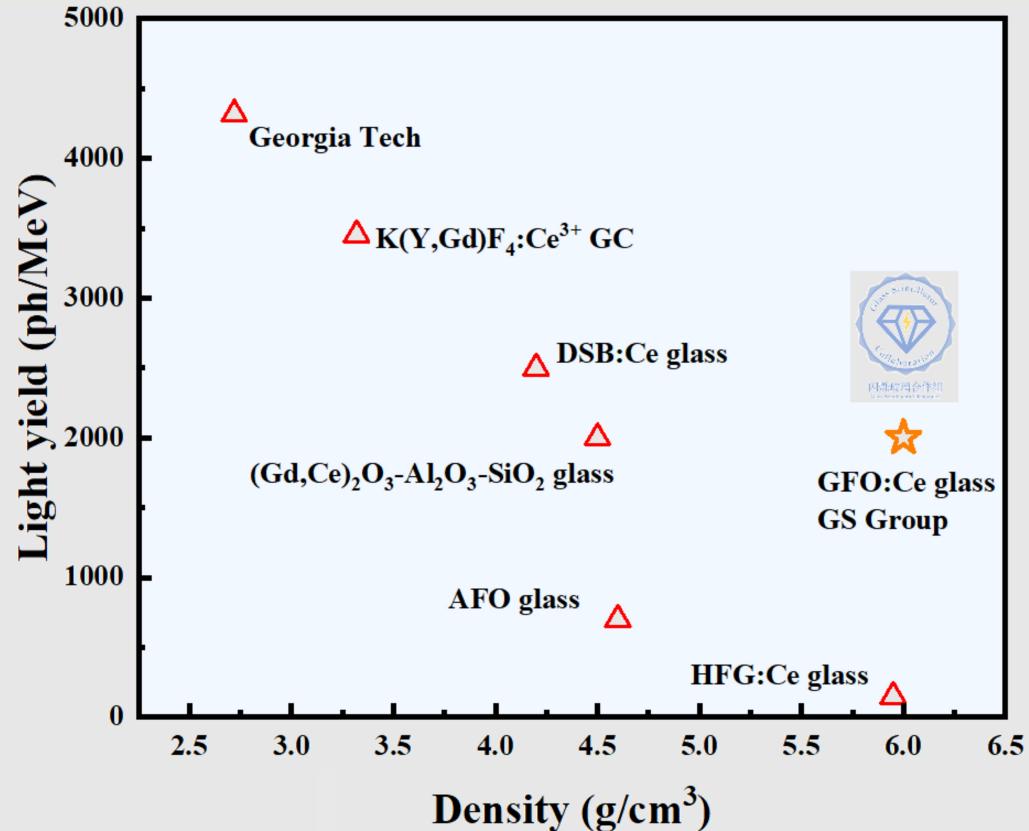
➤ 2023 ACS Appl. Mater. Interfaces.
K(Y,Gd)F₄:Ce³⁺ GC
Density=3.32 g/cm³
LY=3455 ph/MeV

➤ 2025 Ceram. Int.
GFO:Ce glass
Density=6.0 g/cm³
LY=1700 ph/MeV

- Currently, more research on GS is directed towards X-ray imaging, which do not require high-density properties.
- Studies on high-density and light-yield GS remain virtually unexplored internationally.

1.2 Research Progress on GS

1. E. Auffray, et al., *Nucl. Instrum. Methods Phys. Res. A*, 1996, 380, 524-536.
2. V. Dormenev, et al., *Nucl. Instrum. Methods Phys. Res. A*, 2021, 1015, 165762.
3. M.T. Lucchini, et al., *Nucl. Instrum. Methods Phys. Res. A*, 2023, 1051, 168214.
4. A. Amelina, et al., *J. Non-Cryst. Solids*, 2022, 580, 121393.
5. C. Struebing, et al., *J. Am. Ceram. Soc.*, 2017, 101, 1116-1121.
6. C.X. Luo, et al., *ACS Appl. Mater. Interfaces*, 2023, 15, 46226-46235.



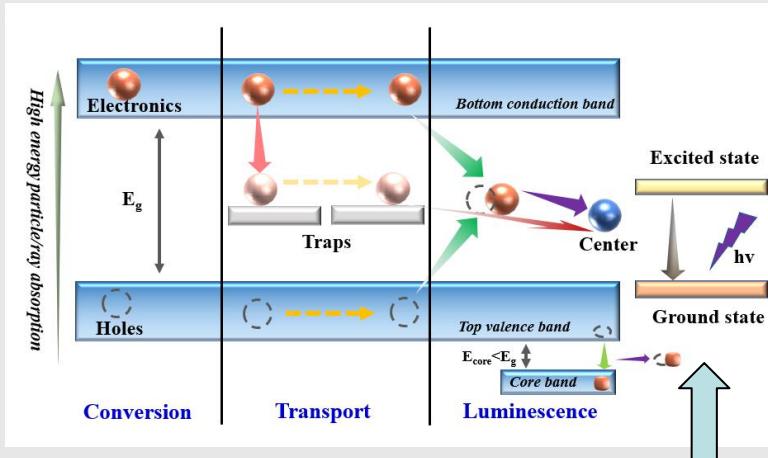
Year	Scintillator	Density (g/cm³)	Light yield (ph/MeV)
1996	HFG:Ce glass	5.95	150
2015	DSB:Ce glass	4.2	2500
2017	Li(Al _y ,Gd _{1-x-y} ,Ce _x)Si(O,F,Br) glass	2.72	4317
2020	AFO glass	4.6	700
2022	(Gd,Ce)₂O₃-Al₂O₃-SiO₂ glass	4.5	2000
2023	K(Y,Gd)F₄:Ce³⁺ GC	3.32	3455
	GFO:Ce glass	6.0	1700

- The introduction of heavy elements and structural densification → high density → more severe self-absorption → difficult to balance density and light yield;
- GFO: Ce is the only type of GS which can achieve both high density and high light yield.

Outline

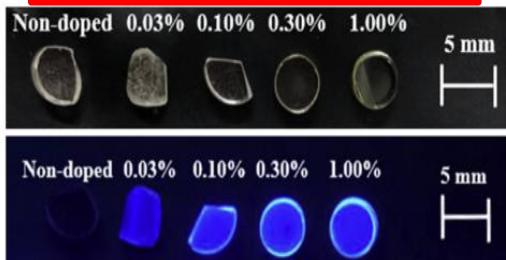
- 1. Performance target of Glass Scintillator (GFO)
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2.0 The Design of the GS



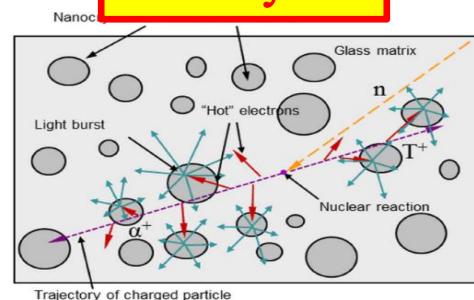
- **Scintillation mechanism**—**Luminescence Center**
- **Conversion**—photoelectric effect and Compton scattering effect;
- **Transport**—electrons and holes migrate;
- **Luminescence**—captured by the luminescent center ions

Lanthanide elements



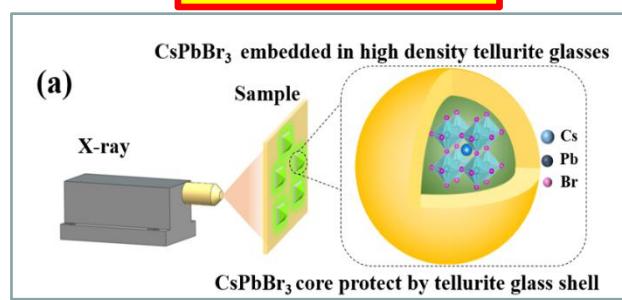
Journal of Alloys and Compounds
782 (2019) 859-864

Nanocrystals



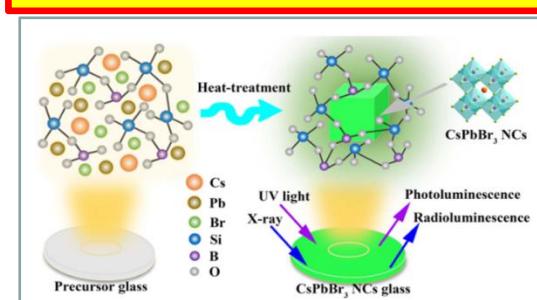
IEEE TNS 60 (2) 2013

Quantum Dots



Optics Letters 46(14) 3448-3451 (2021)

Lanthanide + Quantum Dots



Vol. 9, No. 12 / 2021 / *Photonics Research*

- High Light Yield ($> 2000 \text{ ph/MeV}$): Lanthanide for the Luminescence Center: **Cerium (Ce)**;
- High Density ($> 6 \text{ g/cm}^3$) and Low radioactivity background: **Gadolinium (Gd)**; **Lutetium (Lu)**

2.1 Theoretical feasibility

which then radiate scintillation photons. The light yield LY in terms of the number of scintillation photons produced per MeV of energy deposited in the crystal can be expressed as [64]

$$LY = 10^6 S \cdot Q / (\beta \cdot E_g), \quad (35.3)$$

where $\beta \cdot E_g$ is the energy required to create an e-h pair expressed as a multiple of the band gap energy E_g (eV), S is the efficiency of energy transfer to the luminescent center and Q is the quantum efficiency of the luminescent center. The values of β , S and Q are crystal dependent and

Cite by S.E. Derenzo, W.-S. Choong and W.W. Moses, Phys. Med. Biol. 59, 3261 (2014).

For inorganic **scintillation crystals**, the light yield can be enhanced by improving the intrinsic properties of the material:

1. Reducing the bandgap energy (**Eg**);
2. Improving the energy transfer efficiency (**S**) from the sensitizer to luminescent center (activator) ;
3. Increasing the quantum yield (**Q**) of the luminescent center.

2.2 Enhance in light output

1. Z.H. Hua, et al., *Nucl. Instrum. Methods Phys. Res. A*, 2025, 1072, 170182.
2. Z.H. Hua, et al., *Ceram. Int.*, 2025, 51, 23367-23373.

$$LY = LY_0 \times e^{-\frac{L}{LAL}}, \quad LY = \frac{LY_0 \times (1 - e^{-2\mu L})}{2\mu L}$$

LY_0 —Intrinsic light yield

L—The propagation distance of scintillation light

LAL—Light attenuation length

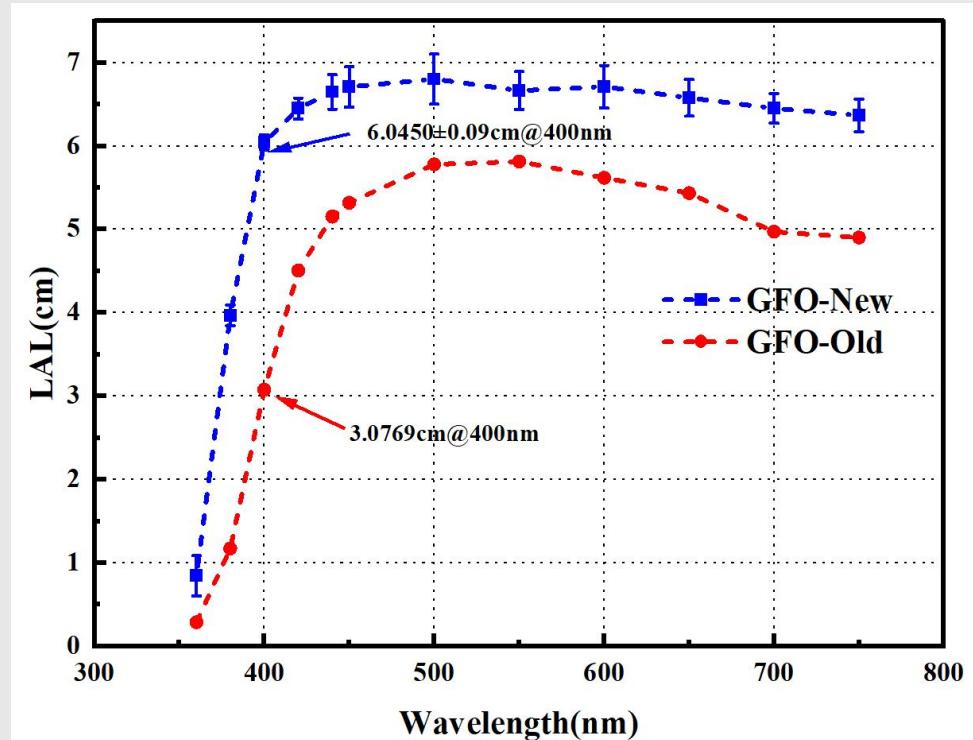
μ —Light loss coefficient (Reabsorption and reflection)

- Reducing losses during scintillation light transmission
- ✓ Improving optical uniformity:
material purity, melting process (bubbles, striae);
- ✓ Enhancing light collection efficiency:
micro-nano structures, reflector selection, coupler;
- ✓ Wavelength shifting:
improving PDE of photoelectric detector.

➤ 2025 Ceram. Int. Study on size effect

➤ 2025 NIMA. Attenuation Length Test

2.3 LAL test

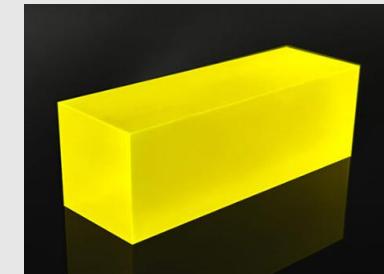


$$I = I_0 \cdot e^{-\frac{l}{L_0}} \quad \rightarrow \quad L_0 = \frac{l_2 - l_1}{\ln(\frac{T_{l_1}}{T_{l_2}})}$$

- Due to its high density and limited homogeneity, glass exhibits strong self-absorption, leading to a relatively short attenuation length.
- The uniformity of glass is related to the purity of the raw materials and the preparation process, which is challenging to control with lab-scale samples.

2.4 GFO:Ce glass to GAGG and GSO crystal

Parameters	Unit	GAGG	GSO	GFO:Ce
Density	g/cm ³	6.63	6.71	6.0
Melting point	°C	1850	1950	1150
Radiation Length, X ₀	cm	1.63	1.38	1.64
Molière radius	cm	2.20	2.96	2.50
Nuclear interaction length	cm	21.5	20.6	24.1
Z _{eff}	--	50.6	58.0	56.9
dE/dX	MeV/cm	8.96	8.5	8.0
Emission peak	nm	540	430	390
Refractive Index	--	1.92	1.85	1.76
Light yield, LY	ph/MeV	60,000	8,000	~ 1,500
Energy resolution, ER	%662keV	5.0	9.2	25.8
Decay time	ns	100, 600	28, 260	101, 1456
d(LY)/dT	%/°C	/	/	/
Cost	\$/cc	/	/	<1.0



GAGG



GSO

- $\text{Gd}_3(\text{Al}_x \text{Ga}_{1-x})_5\text{O}_{12}:\text{Ce}$ (GAGG)
- $\text{Gd}_2\text{SiO}_5:\text{Ce}$ (GSO)
- GFO:Ce Glass

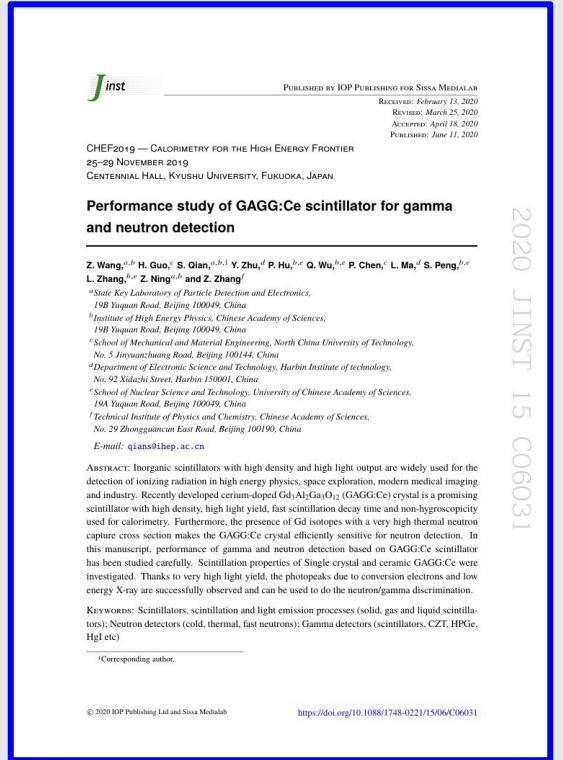
have the similar compositions;
represent the ultimate performance goal.

1. *Y. Zhu, et al., Opt. Mater., 2020, 105, 109964.*
2. *Z. Wang, et al., JINST, 2020, 15, C06031.*
3. *W. Chewpraditkul, et al., Radiat. Meas., 2016, 92, 49-53.*
4. *B.V. Grinyov, et al., IEEE TNS, 2010, 57, 1236-1240.*

GAGG

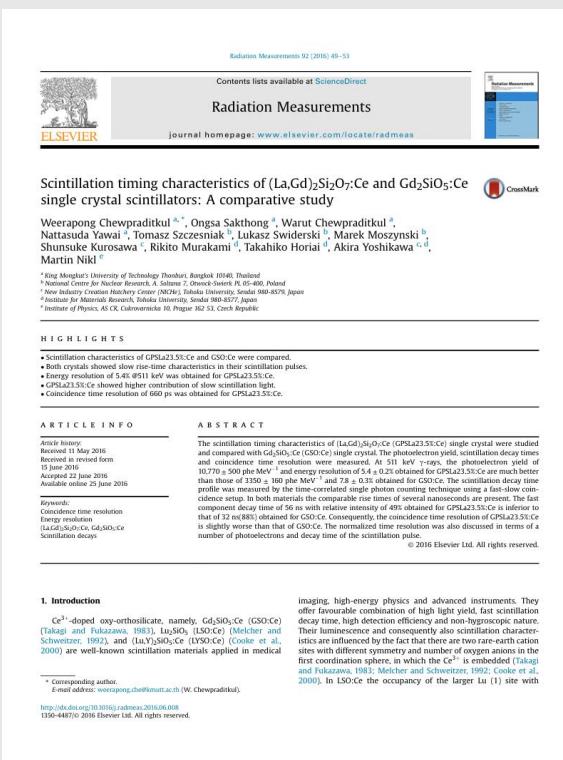


GAGG



➤ 2020 Opt. Mater.
LY=28244±10 ph/MeV
Decay= 50.1, 321.5 ns (60.8%).

GSO



➤ 2016 Radiat. Meas.
LY=3350±160 phe/MeV
Decay= 32, 220 ns

GSO

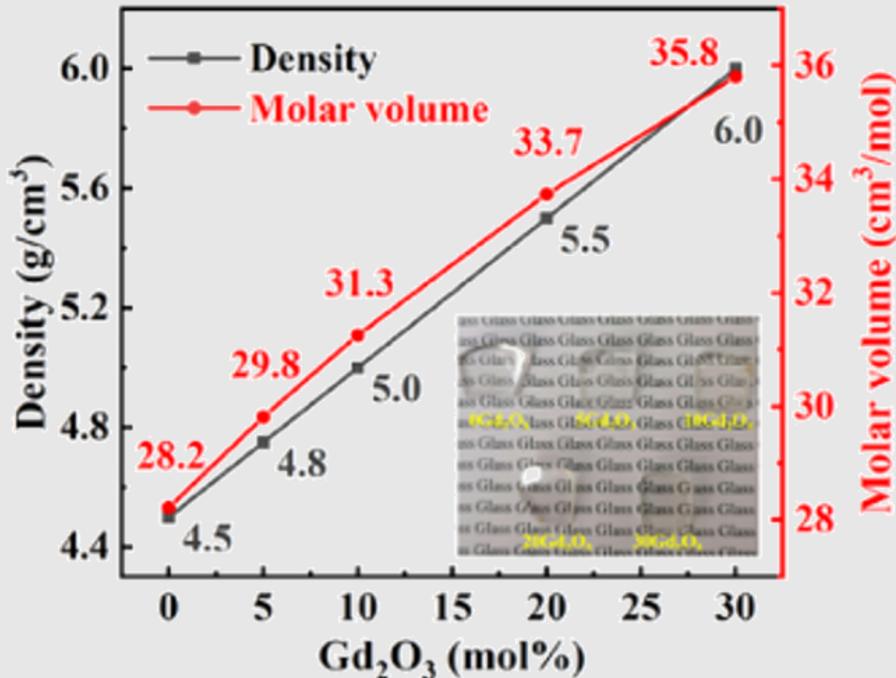


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3.1 Density— (1) Gd_2O_3

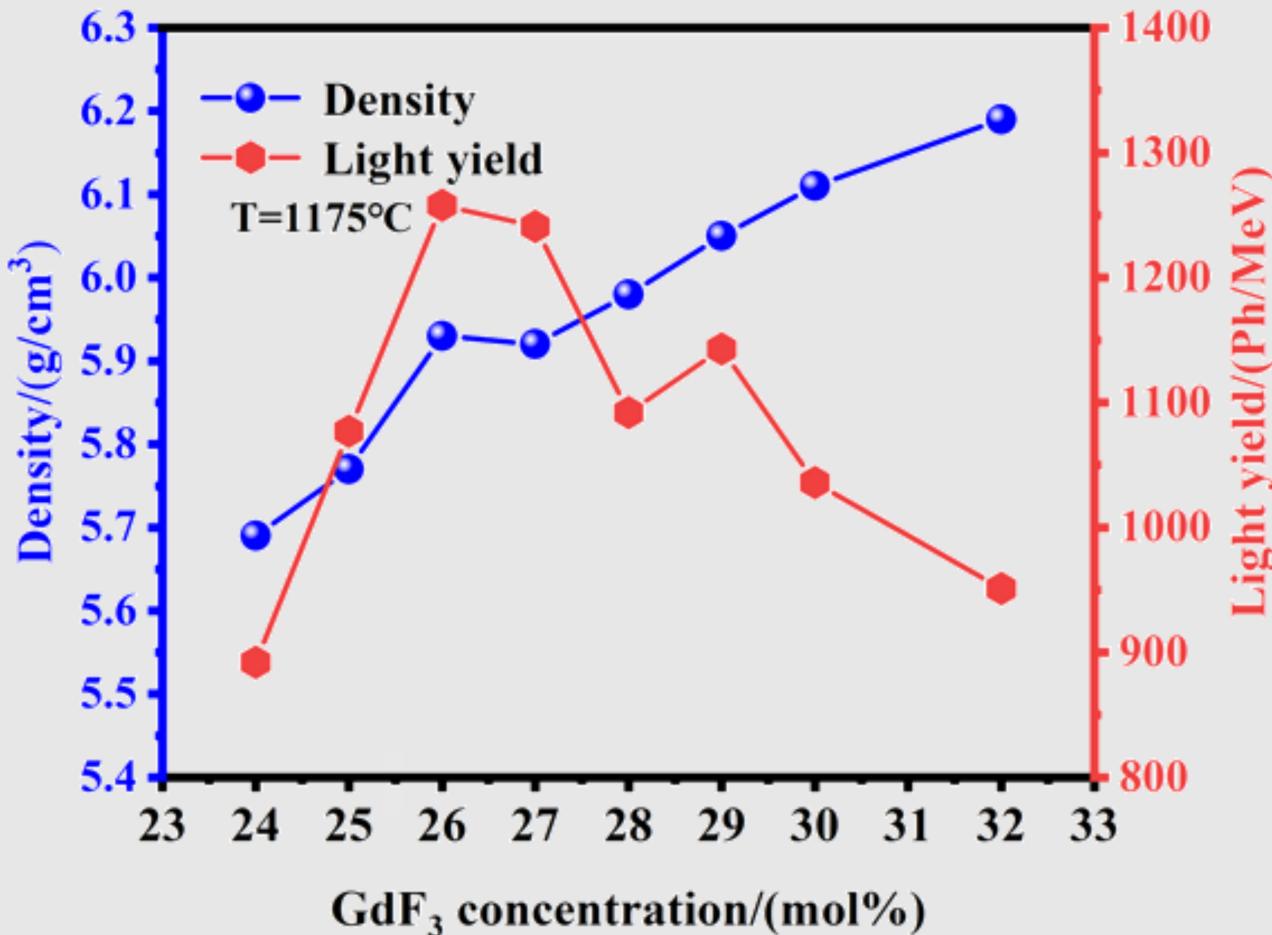
Z.X. Sui, et al., *Ceram. Int.*, 2025, 50, 42625-42631.



	Density (g/cm³)	LY (ph/MeV)
0 Gd_2O_3	4.5	1669
5 Gd_2O_3	4.8	1617
10 Gd_2O_3	5.0	1547
20 Gd_2O_3	5.5	1400
30 Gd_2O_3	6.0	1051

- As the Gd_2O_3 content increases, the glass density rises to 6.0 g/cm^3 ;
- While its light yield declines from around 1700 to $\sim 1000 \text{ ph/MeV}$.

3.1 Density— (2) GdF_3



Recent research progress:

1. By increasing the GdF_3 content, the energy transfer efficiency from Gd to Ce has been enhanced, leading to an improvement in light yield;
2. However, excessive GdF_3 can create quenching centers, which degrade the scintillation performance;
3. By adjusting the ratio of Gd_2O_3 to GdF_3 , the coordination structure of GS has been optimized, resulting in a further increase in light yield.

We really need the density of GFO larger than 6g/cc? What about 5.5? or others?

3.2 Decay Time —GAGG: Ce,Mg crystal

Journal of Crystal Growth 393 (2014) 134–137

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Journal of Crystal Growth

journal homepage: www.elsevier.com/locate/jcrysgrd

Czochralski growth of $\text{Gd}_3(\text{Al}_{5-x}\text{Ga}_x)\text{O}_{12}$ (GAGG) single crystals and their scintillation properties

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B2 Gallium compounds
B3 Oxides
B2 Scintillator materials
B3 Scintillators

ABSTRACT

$\text{Ce:Gd}_3(\text{Al}_{5-x}\text{Ga}_x)\text{O}_{12}$ ($x=2.5/5$ and $3/5$, Ce:GAGG-2.5 and Ce:GAGG-3) crystals were grown by the Czochralski process in order to reduce cost of the starting materials as compared with conventional $\text{Ce:Gd}_3\text{Al}_5\text{O}_{12}$ (Ce:GAGG-2) crystal which have high light output. Although perovskite phase was detected in Ce:GAGG-3, Ce:GAGG-2.5 had single-phase garnet structure. The optical properties including transparency, emission, and excitation spectra of 30 samples cut from the Ce:GAGG-2.5 bulk ingot did not depend on their original position along the growth axis. These samples had light outputs of approximately 58,000 photons/MeV. However, scintillation decay times varied from 140 to 200 ns and depended on the position clearly.

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1. Introduction

Many oxide scintillators such as $\text{Ce}(\text{Lu},\text{Y})\text{SiO}_5$ (Ce:YSO) and $\text{Bi}_2\text{Ge}_3\text{O}_12$ (BGO) are not hygroscopic and demonstrate gamma-ray stopping power greater than that of $\text{Ce}(\text{Lu},\text{Y})\text{SiO}_5$ and $\text{Ce}(\text{Lu},\text{Y})\text{O}_4$. On the other hand, some corresponding oxide crystals have low light output and energy resolution when compared with the halides. However, Ce-doped $\text{Al}_2\text{Ga}_5\text{O}_12$ scintillator has good light output of 46,000 photons/MeV [8]. It is non-hygroscopic, and it has high stopping power exceeding that observed in the halide group. Thus, the Ce-doped $\text{Al}_2\text{Ga}_5\text{O}_12$ crystals are expected to be used in various highly demanded applications including radiation monitoring and medical imaging. Following these circumstances, development of the Czochralski (Cz) process for growth of this type crystal is required to establish their mass production.

Ce:Gd₃(Al₅-_xGa_x)O₁₂ ($x=2.5/5$) crystal that was grown by the Cz process. Other crystals having different ratios and grown by the Cz process have not been yet reported. Ga_{0.5} is more expensive than the Cz process. Therefore, purpose of the current project was to estimate maximum substitution of Ga with Al achievable in the Ce:Gd₃(Al₅-_xGa_x)O₁₂ crystals to reduce cost of the starting materials and the crystals. Moreover, the effective

atomic number of $\text{Ce:Gd}_3(\text{Al}_5\text{Ga}_{1-x})\text{O}_{12}$ ($2/5 < x < 4/5$) is similar to that of $\text{Ce:Gd}_3\text{Al}_5\text{O}_{12}$, and gamma-ray stopping power is also similar. In this report, growth of these crystals with $x=2.5/5$ and $3/5$ using the Cz process is demonstrated together with evaluation of their optical and scintillation properties such as light outputs and scintillation decay times.

2. Materials and experimental methods

The $(\text{Ce}_{0.95}\text{Gd}_{0.05})\text{Al}_2\text{Ga}_{5-x}\text{O}_{12}$ crystals ($x=2.5/5$ and $3/5$) were grown by the Cz process from the melt produced from 99.99% pure starting oxides of $\alpha\text{-Al}_2\text{O}_3$, $\beta\text{-Ca}_3\text{O}_2$, Gd_2O_3 , and Ce_2O_3 . The growth was performed from Ir-crucible on the seed made of $\text{Gd}_2\text{Al}_5\text{Ga}_5\text{O}_{12}$ single crystal. The pulling and rotation rates were $0.1897\text{ mm}/\text{h}$ and 12 rpm , respectively. The growing atmosphere was $\text{Ar}/98\% + \text{O}_2/2\%$ mixture.

The chemical composition of the as grown crystals and distribution of their constituents along and perpendicular to the growth axis were examined using the electron probe microanalysis (EPMA) (JXA-8621MX, JEOL). The measurements were made under accelerating voltage of 20 kV and current of $1.2 \times 10^{-6}\text{ A}$.

The transmittance and reflectance spectra were recorded with spectrophotometers: V-530 (JASCO) and UV-2550 (SHIMADZU), respectively. Photo-luminescence (PL) spectra at room temperature were examined with a spectrophotoflurometer (FL5920, Edinburgh instrument) using a Xenon lamp as an excitation source. Moreover,

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Materials Advances

PAPER

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Compositional engineering of multicomponent garnet scintillators: towards an ultra-accelerated scintillation response[†]

Loris Martinazzoli,^{a,b} Saulius Nargelis,^{a,c} Pavel Boháček,^d Roberto Cala,^{ab} Michal Dušek,^a Jan Rohliček,^a Gintautas Tamulaitis,^a Etiennette Guigay^a and Martin Nikl^{a,*e}

Optical, luminescence and scintillation characteristics were studied in garnet-type GAGG single-crystal scintillators grown by the Czochralski method and heavily doped with a cerium activator and a magnesium codopant at different concentrations. Emission quenching due to the formation of closely spaced Ce–Mg pairs accelerating the photoluminescence and scintillation decays down to a few nanoseconds and substantial suppression of slower decay components are observed. We show that despite a significant decrease in the scintillation yield, the coincidence time resolution and the afterglow, which are the most critically important parameters of fast scintillators, exhibited by the heavily doped GAGG–Ce/Mg are superior to those in the state-of-the-art scintillators. Due to the peculiar feature of the GAGG host to tolerate extremely high cerium and magnesium concentrations while still maintaining a bulk single crystal form, this scintillator has a great potential for high-count-rate applications in high energy physics experiments and industries with harsh operational environments, where a lower light yield can be tolerated.

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rsc.li/materials-advances

1. Introduction

Single crystal scintillators of general formula $(\text{Gd}_x\text{Lu}_y\text{Y}_z)\text{Al}_1\text{Ga}_x\text{O}_3\text{Ce}$, usually referred to as multicomponent garnets, were reported for the first time in combinatorial studies in 2011^{1,2} and immediately gained the immense interest of the scintillator research community due to their high light yield exceeding up to three times that of classical single-crystal scintillators, such as $\text{Y}_3\text{Al}_5\text{O}_13\text{Ce}$ (YAG:Ce) or $\text{Lu}_3\text{Al}_5\text{O}_13\text{Ce}$ (LuAG:Ce) and approaching theoretical limits³ (see ref. 4 for a review). The growth of larger single crystals enabled by the Czochralski method³ is another practical advantage of the multicomponent garnet-type scintillators, though the necessity of using an iridium crucible due to the presence of Ga in the

host composition makes their manufacturing more expensive.⁴ The host with composition $\text{Gd}_x\text{Ga}_x\text{Al}_1\text{O}_{12}$ ($x = 2.5\text{--}3$) became the most intensively studied one, and this Ce-doped scintillator is usually referred to as GAGG:Ce in the literature. The success and origin of two- to three-fold increase in the light yield of this solid solution are based on the ability to engineer the electronic band structure of this material: the addition of Ga lowers the bottom of the conduction band immersing shallow electron traps in the lowered band edge,^{7,8} whereas the Gd admixture ensures a sufficient but energy barrier to prevent the ionization of the Ce^{3+} excited state $5d_1$.^{9–11} It is, however, worth noting that GAGG:Ce scintillators cannot be used above room temperature, because the barrier becomes insufficient to prevent ionization at elevated temperatures,¹² similar to LSO:Ce .¹³ Another positive contribution to the scintillation yield is caused by atomistic inhomogeneities in the cationic sublattice arrangement in GAGG, where local variations in the Ga content induce variations in the bottom of the conduction band. Such a “wavy” shape of the conduction band bottom limits the out-diffusion of electrons from the ionization track and, consequently, enhances their fast radiative recombination with holes captured at cerium ions.¹⁴

An essential step towards further garnet scintillator optimization was accomplished in 2014, when Mg^{2+} -codoping was shown to accelerate the scintillation decay and (for low Mg

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[†] Electronic supplementary information (ESI) available: Compositional engineering of multicomponent garnet scintillators ref. See DOI: <https://doi.org/10.1039/d2ma00626j>

Sample	Light output (ph MeV ⁻¹)	Commercial sample	Light output (ph MeV ⁻¹)
PL1	12 000 \pm 600	GFAG C&A	20 500 \pm 1000
PL2	7600 \pm 380	GAGG Fomos	28 300 \pm 1400
PL3	5200 \pm 260	GAGG ILM	21 000 \pm 1100
PL4	3900 \pm 190		
PL5	2300 \pm 230		
PL6	500 \pm 300		

Sample	τ_r (ps)	τ_{d1} (ns)	R_1 (%)	τ_{d2} (ns)	R_2 (%)	τ_{d3} (ns)	R_3 (%)	$\tau_{d,eff}$ (ns)
C&A GFAG	32	6.0	4.6	44.5	69.2	222	26.3	41
ILM GAGG	37	4.0	3.2	40.4	56.4	138	40.4	40
Fomos GAGG	30	2.2	0.5	53.1	41.7	166	57.8	73
PL1	13	2.5	3.3	25.4	48.0	79.2	48.8	26
PL2	8	2.1	7.2	16.6	54.6	66.2	38.2	13.8
PL3	5	1.6	6.2	12.6	47.5	46.0	46.3	11.6
PL4	5	1.5	9.2	11.3	53.9	45.4	36.9	8.6
PL5	5	1.0	11.0	7.1	51.8	40.8	37.2	5.2
PL6	5	0.2	19.5	1.5	53.0	14.9	27.5	0.7

- LY=58000±3000 ph/MeV
- Decay=200 ns
- LY=500±300 ph/MeV
- Decay=0.2, 1.5, 14.9 ns

3.2 Decay—Ce doping concentration

X.Y. Sun, et al., Chin. Opt. Lett., 2025, 23(12), 121602.

**CHINESE
OPTICS
LETTERS**

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Ultra-high concentration Ce³⁺-doped gadolinium-based borosilicate glass scintillators

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Ce³⁺-doped gadolinium-based borosilicate (GBC₂) glass scintillators with an ultra-high concentration of 16 mol% were synthesized in ambient atmosphere for future calorimeter application. The valence state of Ce was previously controlled in the glass by the X-ray absorption near edge structure (XANES) spectrum. With the increased Ce³⁺ concentration, the NBO/BO ratio decreases notably from 5.15 to 0.56. The GBC₂ glass scintillators exhibit the broad photoluminescence (PL) band within 350–550 nm regions, with a maximum PL quantum yield (PL QY) of 60.8%. In X-ray excited luminescence (XEL), the integral intensity of the GBC₂ glass is 18.4% compared to the BG0 crystal. Meanwhile, it has the highest light yield of 1043 ph/MeV with an energy resolution of 28.4% at 662 keV under γ -ray excitation. When the doped concentration of Ce³⁺ exceeds 4 mol%, the proportion of light yield within 1 μ s integral gate exceeds 95%, which conforms to the requirement of fast time response. Interestingly, the concentration quenching effect of high concentration Ce³⁺ ($x \leq 14$) does not occur in the glass scintillators under γ -ray excitation. With the increase of Ce³⁺ concentration, both the fast (100–18 ns) and slow (1000–59 ns) components of scintillation decay time decrease dramatically. Therefore, the developed GBC₂ glass scintillators, featured with the reasonable light yield and fast time response, have a promising application in future high energy physics (HEP) experiments.

Keywords: HEP experiment; glass scintillator; XANES; light yield; fast scintillation decay.

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1. Introduction

In recent decades, the purpose of scintillation materials for detectors and calorimeter systems has been a major factor in particle physics, nuclear medicine, and even astrophysics fields^[1–3]. Currently, glass scintillators are targeted as static

X-ray imaging^[4–6], as they only pursue higher light yield with less care for scintillation time and glass density. However, HEP experiments will require a comprehensive advantage of scintillation materials, including low cost, high density, adequate energy resolution, and fast decay. It is still difficult to achieve

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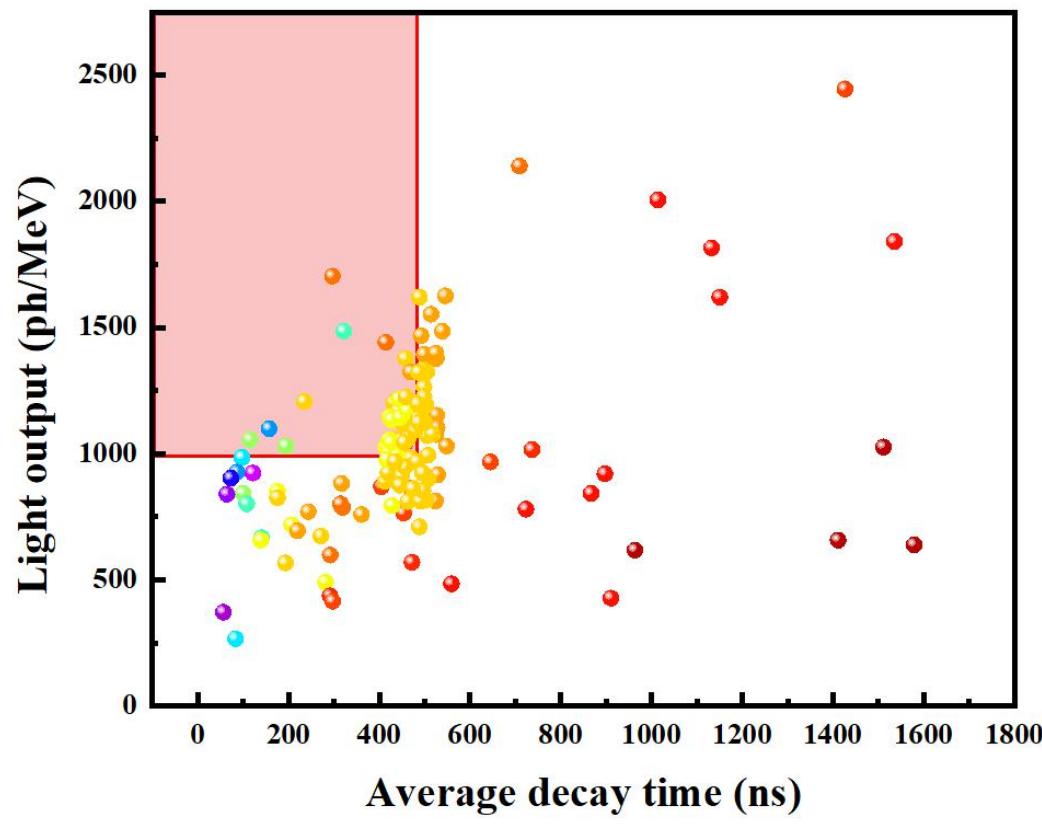
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	Density (g/cm ³)	Light yield (ph/MeV)	Scintillation decay time (ns)
GFO1	6.1400	427	98.9 (1.6%), 924.2
GFO2	6.0494	1043	98.6 (8.2%), 490.7
GFO5	5.9736	825	82.0 (27.5%), 211.5
GFO9	5.9332	985	35.8 (10.7%), 105.1
GFO14	5.9152	839	25.3 (10.3%), 69.0
GFO16	5.8917	371	18.9 (7.2%), 59.7

- Optimizing the decay time of GS still relies on adjusting the concentration of Ce³⁺ centers.
- As the Ce doping concentration increases, the scintillation decay time gradually decreases.
- However, the light output first increases and then decreases.
- It is necessary to balance light yield and decay time.

Relationship between decay and light output



- Over the past four years, the GS group has prepared more than 1500 pieces of glass;
- Decay time of the GS, determined by physical properties: 100ns -> 300ns -> 500ns;
- Typical results show that decay time and light yield tend to be positively correlated: a higher light yield generally implies a slower decay time;
- Only a small number of GS samples meet all three performance criteria: density $> 6.0 \text{ g/cm}^3$, light yield $> 1000 \text{ ph/MeV}$, and decay time $< 500 \text{ ns}$.

We really need the decay time of GFO less than 500ns? What about 1us? or others?

3.3 Peak emission wavelength shift

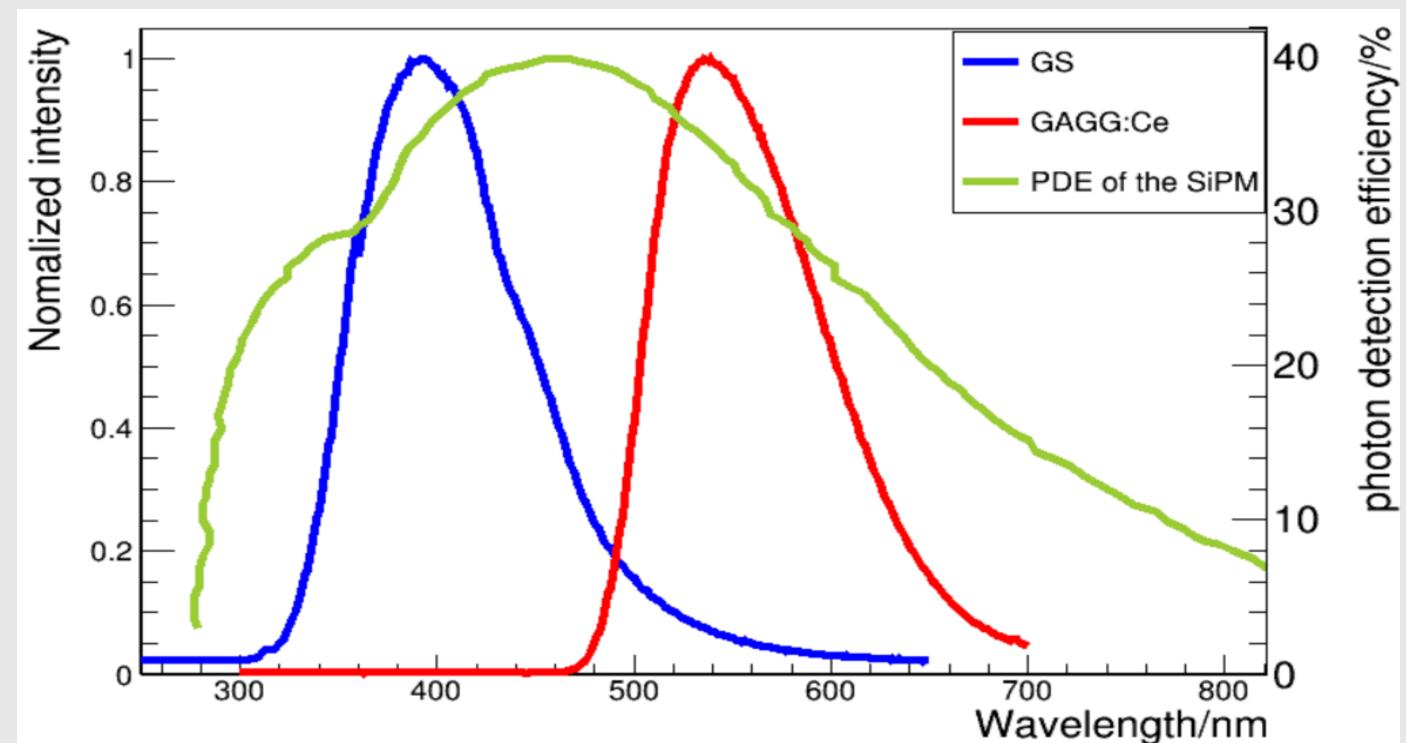
light yield (*LY*) in number of photons/MeV produced ($N_{\text{photons}}/\text{MeV}$) and the light output (*LO*) in number of photoelectrons/MeV detected involves the factors for the light collection efficiency (*LCE*) and the quantum efficiency (*QE*) of the photodetector:

$$LO = LY \cdot LCE \cdot QE. \quad (35.4)$$

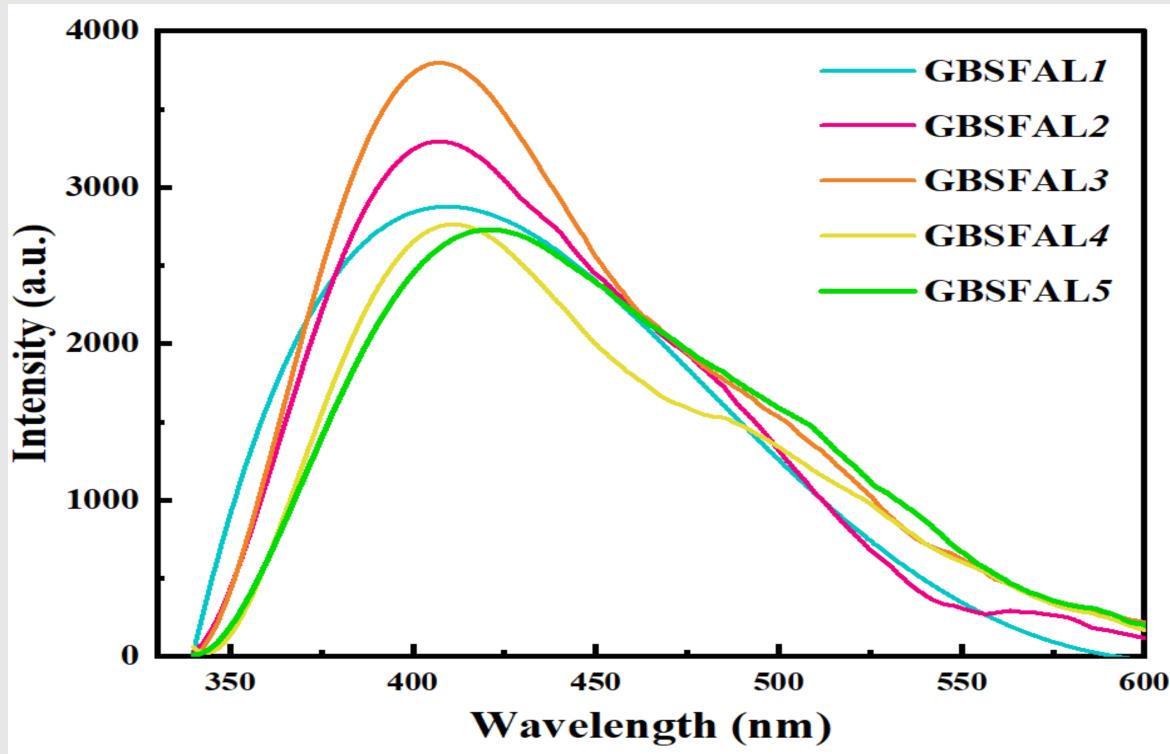
Cite by R.H. Mao, L.Y. Zhang and R.-Y. Zhu, IEEE Trans.NS55, 2425 (2008).

Enhancing the measured light output:

- Improving light collection efficiency
- Matching with the QE of photodetectors



3.3 Wavelength shift of GS (Unpublished)



Doping **X** into GFO Glass

	XEL peak (nm)	LO (ph/MeV)	ER (% @662keV)
GBSFAL1	407	1271	29.8
GBSFAL2	408	1360	26.3
GBSFAL3	409	1491	26.9
GBSFAL4	411	1116	29.7
GBSFAL5	420	1252	28.1

- After adding **X⁺**, the glass XEL peak red-shifts from 390 nm to approximately 420 nm;
- As a strong network modifier, **X⁺** increases the coordination field around Ce³⁺, lowers its luminescence energy levels, and ultimately leads to a redshift of the XEL peak and an increase in light output;
- However, excessive **X⁺** can disrupt the glass network structure and reduce the light yield.

3.4 Ion co-doping— $\text{Cs}_3\text{Cu}_2\text{I}_5:\text{Ti}$ (CCI crystal)

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RAPID RESEARCH LETTER



Zero-Dimensional $\text{Cs}_3\text{Cu}_2\text{I}_5$ Perovskite Single Crystal as Sensitive X-Ray and γ -Ray Scintillator

Shuangliang Cheng, Alena Beiterlova, Romana Kucerkova, Martin Nikl, Guohao Ren, and Yuntao Wu*

Low-dimensional halide perovskites with highly efficient luminescent properties have attracted huge attention in optoelectronic and radiation detection applications in recent years. Herein, a highly efficient 0D $\text{Cs}_3\text{Cu}_2\text{I}_5$ perovskite crystal is presented for X-ray and γ -ray detection that can be easily grown from the melt by the Bridgman method. The crystal structure of $\text{Cs}_3\text{Cu}_2\text{I}_5$ belongs to an orthorhombic system with a space group of Pmm . The $\text{Cs}_3\text{Cu}_2\text{I}_5$ crystal is nonhygroscopic and self-absorption free due to a large Stokes shift of 120 nm originated from its self-trapped exciton emission. The scintillation emission centers at 440 nm with a principal decay time of 967 ns. The $\text{Cs}_3\text{Cu}_2\text{I}_5$ crystal not only demonstrates a high scintillation yield of about 32 000 photons MeV⁻¹ and an extremely low afterglow of 0.03% at 10 ms under X-ray radiation, but also possesses a high light yield of 29 000 photons MeV⁻¹ with an excellent energy resolution of 3.4% at 662 keV under γ -ray radiation.

Organic-inorganic hybrid halide perovskites, represented by $\text{CH}_3\text{NH}_3\text{PbX}_3$ (X = Cl, Br, I), have demonstrated excellent optoelectronic and radiation detection properties.^[1-3] All-inorganic CsPbX_3 perovskites (X = Cl, Br, I) with better long-term stability are also considered as promising materials for optoelectronic devices and semiconductor γ -ray detectors.^[4-6] Recently, there has been a surge of interest in low-dimensional perovskites due to their high photoluminescence quantum yield (PLQY). For example, the crystal structure of $\text{Cs}_3\text{Cu}_2\text{I}_5$, that consist of spatially isolated $[\text{PbBr}_4]^{4-}$ octahedra surrounding with Cs^+ ions can be regarded as a 0D structure at atomic level, which leads to an intense quantum confinement effect. Excitons are strongly confined at each $[\text{PbBr}_4]^{4-}$ octahedron, enabling a high exciton

as sensitive X-ray scintillators, such as Rb_2CuBr_4 and $\text{Rb}_3\text{Cu}_2\text{I}_5$.^[7-11] In particular, the Rb_2CuBr_4 , that is self-absorption free and nonhygroscopic, was reported to achieve an ultrahigh scintillation yield of 90 000 photons MeV⁻¹.^[12]

All-inorganic 0D perovskite $\text{Cs}_3\text{Cu}_2\text{I}_5$ was recently reported as a highly efficient blue-emitting material with a PLQY of 91.2%, and regarded as promising for application in photodetectors, light-emitting diodes, and memristors afterward.^[14-17] In 2020, $\text{Cs}_3\text{Cu}_2\text{I}_5$ nanocrystals were developed for X-ray imaging with a light yield of 80 000 photons MeV⁻¹.^[18] To the best of our knowledge, the X-ray and γ -ray detection capability of bulk $\text{Cs}_3\text{Cu}_2\text{I}_5$ single crystal has not been reported. Thus, the aim of this work is to study the physical and optical properties and the scintillation performance under X-ray and γ -ray radiation of high-dimensional $\text{Cs}_3\text{Cu}_2\text{I}_5$ perovskite single crystal grown by the Bridgman method.

The 7-mm diameter single crystal of $\text{Cs}_3\text{Cu}_2\text{I}_5$ was grown by the vertical Bridgman method. High-purity powders of CsI (99.99%, Grivin Advanced Materials) and CuI (99.999%, Sigma-Aldrich) were used as raw materials. These starting materials were mixed consistent with stoichiometric ratio and loaded into a quartz ampoule in a glovebox with $<0.1\text{ ppm}$ moisture and oxygen. The loaded ampoule was sealed under a vacuum of 10^{-6} torr after drying at 180 °C for 16 h, and then placed into a Bridgman growth furnace. Prior to growth, the $\text{Cs}_3\text{Cu}_2\text{I}_5$ compound was synthesized under 630 °C (a few degrees above the melting points of all components) for 24 h. Then the growth process started at 390 °C^[19] with a temperature gradient of 20–30 °C cm⁻¹ and a translation rate of 0.5 mm h⁻¹. The furnace

Add In
Indium

RESEARCH ARTICLE



Highly Resolved X-Ray Imaging Enabled by In(I) Doped Perovskite-Like $\text{Cs}_3\text{Cu}_2\text{I}_5$ Single Crystal Scintillator

Qian Wang, Quan Zhou, Martin Nikl, Jiawen Xiao,* Romana Kucerkova, Alena Beiterlova, Vladimir Babin, Petr Prusa, Vladimir Linhart, Jingkang Wang, Xuemin Wen, Guangda Niu, Jiang Tang, Guohao Ren, and Yuntao Wu*

Low-dimensional perovskite halides have shown a great potential as X-ray detection materials because of efficient exciton emissions originating from strongly spatially localized charge carriers. Nonetheless, most of them have a scintillation yield far below their theoretical limits. Here, it is found that the harvesting efficiency of produced charge carriers can be significantly enhanced via a small amount of In⁺ doping in these highly localized structures. A bright and sensitive zero-dimensional $\text{Cs}_3\text{Cu}_2\text{I}_5:\text{In}^+$ halide with efficient and tunable dual emission is reported. The radio-luminescence emission of $\text{Cs}_3\text{Cu}_2\text{I}_5:\text{In}^+$ crystals under X-ray excitation consists of a self-trapped exciton emission at 460 nm and an In⁺-related emission at 620 nm at room temperature. In⁺ doping enhances the photoluminescence quantum efficiency (PLQY) of $\text{Cs}_3\text{Cu}_2\text{I}_5$ from 68.1% to 88.4%. Benefiting from the higher PLQY, $\text{Cs}_3\text{Cu}_2\text{I}_5:\text{In}^+$ can achieve an excellent X-ray detection limit of 96.2 nR_{Fe} s⁻¹, and a superior scintillation yield of 53 000 photons per MeV, which is comparable to commercial CsI:Tl single crystals. As a result, a remarkable X-ray imaging resolution of 18 line pairs mm⁻¹ is demonstrated, which is so far a record resolution for single crystal perovskite-based flat-panel detectors. These results highlight the importance of efficient harvesting of carriers (and excitons) in low-dimensional perovskites for radiation detection applications.^[20]

1. Introduction

Metal halide perovskites are a promising platform for photovoltaics,^[1] light-emitting applications,^[2] photodetectors,^[3] lasers,^[4] and other optoelectronic applications owing to their extraordinary optoelectronic properties.^[5] In particular, these materials also show great potential in ionizing-radiation high-energy radiation detection due to their high effective atomic number (Z_{eff}) and high luminescence efficiency.^[6] Since the first report on lead halide perovskite nanocrystallin scintillators by Liu's group,^[7] many efforts have been devoted to exploring the application of halide perovskite scintillators for radiation detection.^[8,9] However, though the halide perovskite scintillators are highly promising, the light yield (LY , $\sim 10^4$ –10⁵ photons per MeV) is still far below that of commercial CsI:Tl, such as CsI:Tl (54 000 photons per MeV), Cd₃O₂S (60 000 photons per MeV).^[8,9] Also, the stability of lead halide perovskite is still a great challenge for practical applications.^[10]

Add Tl
Thallium

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2200304 (1 of 8)

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Crystal growth and scintillation properties of pure and Tl-doped $\text{Cs}_3\text{Cu}_2\text{I}_5$

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ARTICLE INFO

ABSTRACT

In this work the Bridgman crystal growth and scintillation properties of both undoped and Tl-doped $\text{Cs}_3\text{Cu}_2\text{I}_5$ are presented. This material is very sensitive gamma and X-ray detector with a density of 4.53 g/cm³ and atomic number of 51.9. Tl-doped $\text{Cs}_3\text{Cu}_2\text{I}_5$ has a light yield of 44 000 photons MeV⁻¹ with an energy resolution of 4.6% at 662 keV. Thallium doping at 0.5 mol % resulted in a band-improved scintillation response, in which light yield increased to 98 200 photons/MeV and energy resolution reduced to 3.3% at 662 keV. The X-ray excited emission is centered at 442 nm for the undoped and 500 nm for the Tl-doped crystals. The undoped emission is broad, typical of excitonic emission, while thallium doping results in an even broader band with features of both the undoped and thallium-doped emission.

1. Introduction

There has been some interest in using this material in optoelectronic devices. Jun, et al. reported a fabrication of a blue light-emitting diode (LED) as well as preparing white luminescence films by mixing $\text{Cs}_3\text{Cu}_2\text{I}_5$ with a yellow phosphor.^[12] Zhang, et al. reported the use of a solvent evaporation crystallization technique to produce millimeter-size single crystals and powders with strong blue photoluminescence and a band gap of 2.0 eV (50% of $\text{Cs}_3\text{Cu}_2\text{I}_5$) and 60% (with the aid of Cs_2CO_3), and a red phosphor with a light yield of 10 000 photons per MeV.^[13] Wu, et al. activated $\text{Cs}_3\text{Cu}_2\text{I}_5$ phosphors have been studied for use in white LEDs.^[14] While a $\text{Cs}_3\text{Cu}_2\text{I}_5$ thin film has been used as a component of a deep ultraviolet photodetector.^[15]

These reports suggest promise for $\text{Cs}_3\text{Cu}_2\text{I}_5$, as does its density of 4.53 g/cm³ and effective atomic number of 51.9 [8]. It has long been established that the performance of CsI is significantly improved by doping with thallium [16–21]. Our hypothesis, the basis for this current work, was that doping $\text{Cs}_3\text{Cu}_2\text{I}_5$ with thallium would result in similarly enhanced performance. In this manuscript we have established that $\text{Cs}_3\text{Cu}_2\text{I}_5$ single crystals are intrinsic scintillators, although they lack the high light yield and excellent energy resolution required for spectroscopy, and that doping $\text{Cs}_3\text{Cu}_2\text{I}_5$ with thallium does indeed result

in the availability of high-performance, low-cost scintillators that has led to ongoing research by many groups in the search for new scintillators [1,2]. At the University of Tennessee, we have an active scintillator discovery program for materials for the detection of ionizing radiation. High density, high scintillation light yield, and good energy resolution are important for gamma-ray spectroscopy applications. In this work, we have focused on a new solid-state scintillator, the alkali copper iodide, specifically $\text{Cs}_3\text{Cu}_2\text{I}_5$. The existence of this material has been known for many years, and some of the properties have been investigated. Bigalk and synthesized $\text{Cs}_3\text{Cu}_2\text{I}_5$ in 1988, using both solid state synthesis and an organic solvent method, and reported that it has an orthorhombic structure with space group $Pbnm$ [3]. Phase diagrams were published by Wojciechowicz, et al. in the same year [4]. The properties of $\text{Cs}_3\text{Cu}_2\text{I}_5$ have been reported [5–8] as having the properties of nanocrystals. Cheng, Lian, and Tao, et al. have all reported that undoped nanocrystals have a self-trapped exciton (STE) emission, photoluminescence and radio-luminescence peaks around 446 nm, and a photoluminescence lifetime of about 1.9 μ s, the bandgap was reported to be 4.60 eV [9–11].

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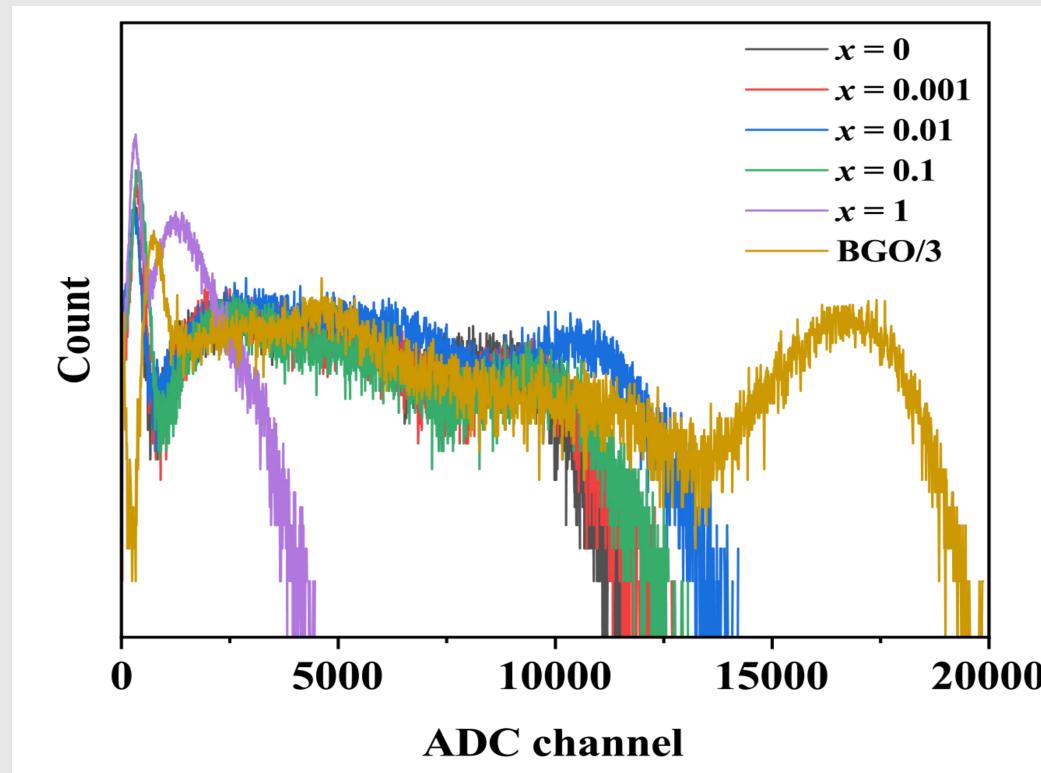
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- LY=28000 photons/MeV
- E.R=3.4%@662KeV
- Decay=967 ns
- LY=44000 photons/MeV
- Decay=3746.5 ns
- LY=98200 photons/MeV
- E.R=3.3%@662KeV
- Decay=840 ns

3.4 GS—co-doped with Ce and Y (Unpublished)



Doping Y into GFO Glass

	LY (ph/MeV)	ER (% @662keV)
1Ce+0Y	1185	40.6
1Ce+0.001Y	1301	33.8
1Ce+0.01Y	1485	31.5
1Ce+0.1Y	1350	34.8
1Ce+1Y	331	75.2

- Incorporating additional luminescent centers can reduce the non-radiative recombination between self-trapped excitons and glass defects, offering a potential approach to enhance the light yield of the GS.
- By adding a trace amount of Y^+ , the light yield increased from 1200 to ~ 1500 ph/MeV.

Summary

- 1. High Light Yield Enhancement is possible!
- 2. We try our best to Enhance the Light Yield;
 - Density: We really need the density of GFO larger than 6g/cc?
 - Decay Time: We really need the decay time of GFO less than 500ns?
 - Wavelength Shift:
 - Co-doping: We need time and funding to find the new materials;
- 3. How to control the cost of GFO Glass.



See the unseen
change the unchanged

THANKS

The Innovation

