



Ultra-low radionuclide concentration measurement lab in School of Physics and High Energy Physics Centre, PKU*

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*Supported by Pandax Group



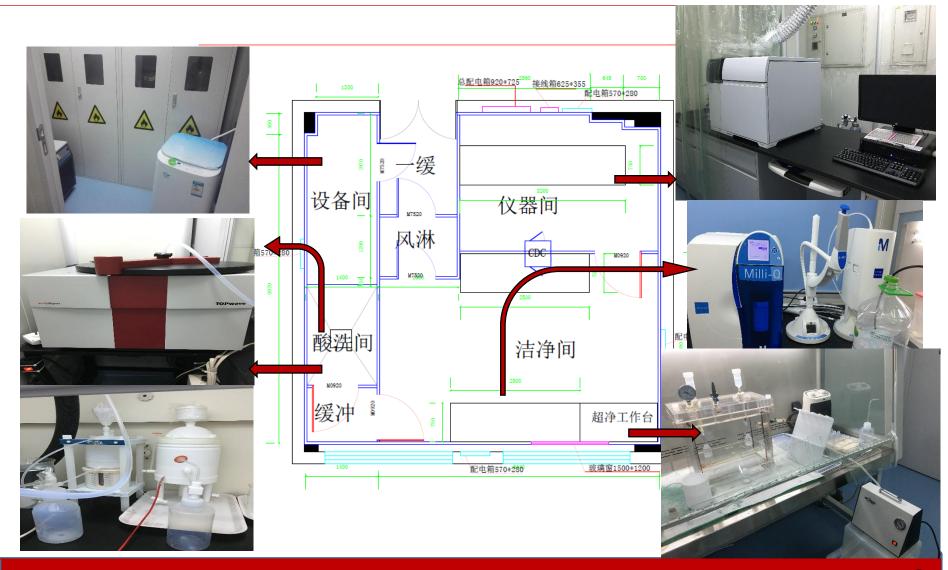
Outline

1. Brief introduction to Trace Analysis Lab

2. ICP-MS & Sample Preparation System

3. Several Exploratory Experiments

Schematic Diagram for the lab



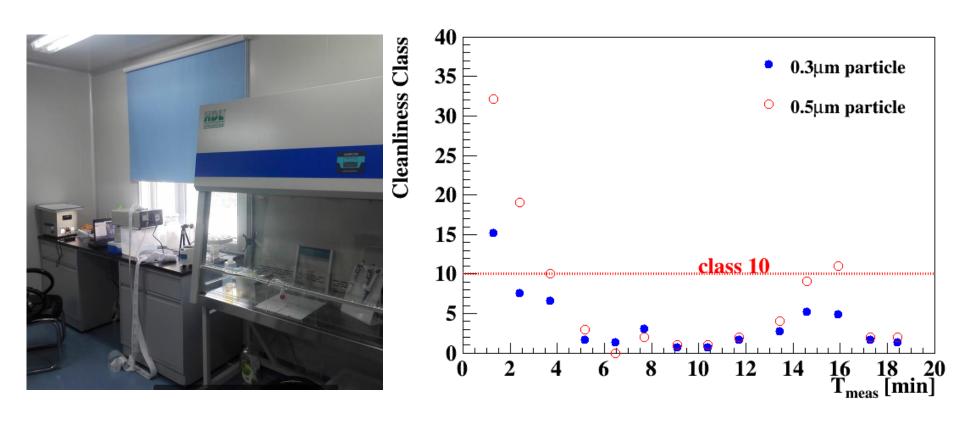
ISO 14644-1 cleanroom standards

Cleanrooms are classified according to the counts and size of particles permitted per volume of air.

Class	maximum particles/m³												
Class	≥0.1 µm	≥0.2 µm	≥0.3 µm	≥0.5 µm	≥1 µm	≥5 µm	equivalent						
ISO 1	10	2.37	1.02	0.35	0.083	0.0029							
ISO 2	100	23.7	10.2	3.5	0.83	0.029							
ISO 3	1,000	237	102	35	8.3	0.29	Class 1						
ISO 4	10,000	2,370	1,020	352	83	2.9	Class 10						
ISO 5	100,000	23,700	10,200	3,520	832	29	Class 100						
ISO 6	1.0×10 ⁶	237,000	102,000	35,200	8,320	293	Class 1,000						
ISO 7	1.0×10^{7}	2.37×10^{6}	1,020,000	352,000	83,200	2,930	Class 10,000						
ISO 8	1.0×10 ⁸	2.37×10^{7}	1.02×10^{7}	3,520,000	832,000	29,300	Class 100,000						
ISO 9	1.0×10 ⁹	2.37×10^{8}	1.02×10^{8}	35,200,000	8,320,000	293,000	Room air						

https://en.wikipedia.org/wiki/Cleanroom

Cleanliness Measurement in Sample Preparation Room (Apr. 16,2016)



Cleanliness in Chemical Hood(Mar. 7, 2016)



Time(mi nutes)	Class (0.3µm)	Class (0.5μm)	Class (1.0µm)
0	104. 6	200.8	344. 0
1.45	27.4	70.3	85.0
2.90	10.4	21. 1	21.3
4.35	3.8	5. 0	12.7
5.80	2.8	1.0	0.0
7. 25	6. 2	11.1	4. 2
8.70	4.9	8.0	4. 2
10. 15	2. 1	4.0	8.5
11.60	2. 1	4.0	4. 2
13.05	4. 2	2.0	0.0
14.50	4. 5	7.0	8.5
15.95	5. 2	6.0	0.0
17. 40	4. 2	6. 0	8.5

Cleanliness Measurement in Super Clean Bench (Apr.26,2016)



ROOM:02 LOCATION: 01/01 CYCLE: 001/001 DATE:2016-04-26 TIME: 10:21:01 PERIOD: 120Sec VOLUME:1m ³ SIZE CUMU 0.3Mm 18 0.5Mm 0 1.0Mm 0	LOCATION: 01/01 CYCLE: 003/001 DATE:2016-04-26 TIME: 10:25:45 PERIOD: 120Sec VOLUME:1m ³ SIZE CUMU 0.3Mm 18 0.5Mm 0 1.0Mm 0 3.0Mm 0
ROOM:02 LOCATION: 01/01 CYCLE: 002/001 DATE:2016-04-26 TIME: 10:23:23 PERIOD: 120Sec VOLUME:1m ³ SIZE CUMU 0.3Mm 0 0.5Mm 0 1.0Mm 0	ROOM:02 LOCATION: 01/01 CYCLE: 004/001 DATE:2016-04-26 TIME: 10:28:07 PERIOD: 120Sec VOLUME:1m ³ SIZE CUMU 0.3Mm 0 0.5Mm 0 1.0Mm 0 3.0Mm 0

16.8.24 7

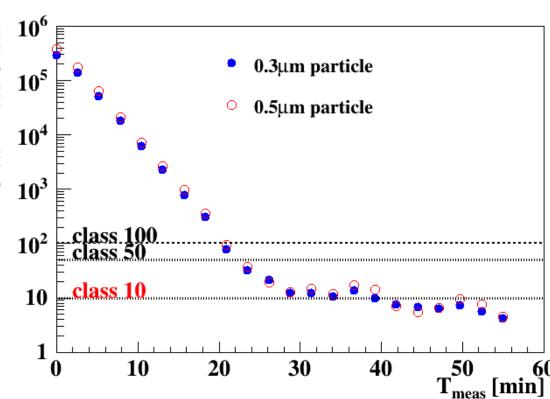
Cleanliness Measurement in the Injection Port of ICP-MS



Time(minut es)	Class (0.3µm)	Class (0.5µm)	Class (1.0μm)
0	0.0	0.0	0.0
1.45	0.0	0.0	0.0
2.9	0.0	0.0	0.0
4.35	0.3	0.0	0.0

Cleanliness Measurement in buffer area (May 8, 2016)





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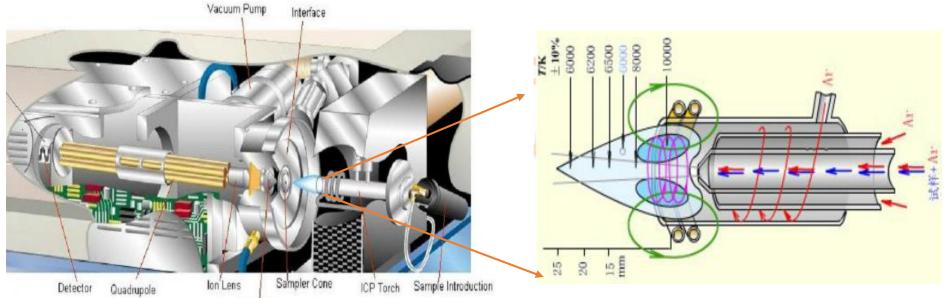
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ICP-MS



- > High frequency current excites alternating electromagnetic field
- Argon gas was ionized by high voltage electric spark, the ions of Argon are collided each other to product inductive current and heat the gas into plasma
- Sample molecules break apart, and then the constituent atoms ionize.
- ➤ Ions are separated on the basis of their mass-charge ratio in quadrupole and a detector receives an ion signal proportional to the concentration

ICP-MS (Agilent 7900)





Agilent Technologies

承诺书

致: 北京大学实验室与设备管理部

关于贵方设备采购项目:北京大学物理学院"电感耦合等离子体 质谱仪"招标采购项目(第2015【022】号),我方作出以下承诺:

- 设备到货后, U238 灵敏度(按照斜率计算)能达到 1000Mcps/ppm,同时,检出限可以达到5ppq;

此承诺。

Agilent Technologies Singapore (Sales) Pte Ltd.

2015年7月21日

CHEONG Chargetus, povid Director, business Operations South Aum Pacific and Korea Regional Sales Operations For and on behalf of AGILENT LECTHOLOGIES SINGLAPORE (SALES) PTE LTD LICENDES ARS MITMEN NO. 10028470176)

ICP-MS (Agilent 7900)

```
232 Th [ No Gas ]
   x_{10} = 998.5694 * x + 6.6000
         检测限 = 0.003179 ppt
    1.5 BEC = 0.006609 ppt
235 [ No Gas ]
        y = 1425.5137 * x + 0.5300
         R = 1.0000
         BEC = 0.0003718 ppt
       [ No Gas ]
238 U
          = 1367.2684 * x + 3.8667
         检测服 = 0.00368 ppt
      2 | BEC = 0.002828 ppt
```

1. Preparation of Reagents



Savillex DST-1000



 \triangleright Experimental water : 18.2M Ω ·cm Milli-Q water

Th & U in HNO₃ and HCl

Sample	232 Th CPS	235 U CPS	238 U CPS
	(RSD)	(RSD)	(RSD)
5% HNO ₃	6.6	0.53	3.867
	(16.035)	(75.472)	(43.378)
0.1 M HCl	3.733	-0.47	0.933
	(24.157)	(0)	(53.927)

2. Preparation of Labware and Sample

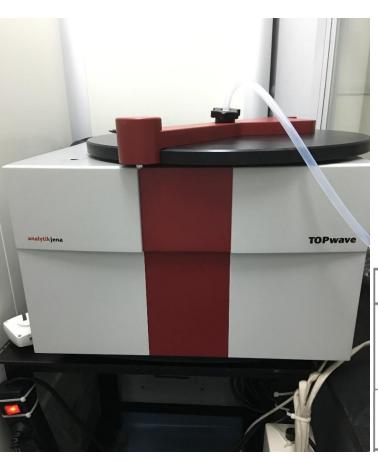


weighting error :0.02mg

- Before using, all vials, comtainers and lids were **soaked in HNO**₃ at least for three days
- After proper leach with 18.2MΩ.cm
 Milli-Q deionized water, the labware
 required were cleaned by ultrasonic
 cleaner in UPW for at least ten minutes

At last , triply rinse with 18.2M Ω .cm Milli-Q deionized water, blow dry and set aside

3. Sample digestion (analytikjena TOPwave)





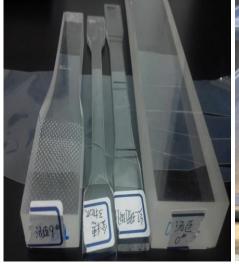


Material	TFM®, P	EEK, quartz, a	luminium oxide		
	Volume	Max. pressure	Max. temperature	max. sample weight (organic)	Min. fill volume (acid)
CX 100	100 ml	100 bar (1450 psi)	230 °C (446 °F)	< 1000 mg	> 7 ml

Microwave System for Pressure Digestion (analytikjena TOPwave)





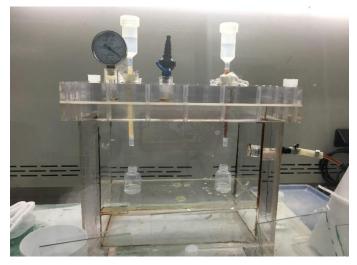


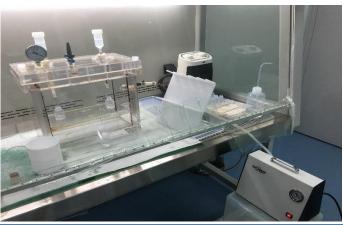


	select as t	avourite (0.	.12): 5		
	temperatu	re pressure	ramp	time	power
1	90	80	5	10	60
2	120	80	5	10	80
3	220	80	10	30	80
4	50	80	5	5	0
5	50	80	5	0	0
int	p-band	total time	vessel	mag	gT
1	20	85	8		120

	T	p	ramp	time	power			
1	110	80	80 20 5					
2	170	80	90					
3	200	80	15	15	90			
4	230	80	5	20	90			
5	50	0	1	1	0			
int	p-band	total time	vesse	1 1	nag temp.			
20	20	102	8		120			

4. Separation & Preconcentration





Anion exchange resin:

Bio-Rad Corp AG 1×4, 100–200

Separation:

$$2R-NO_3^- + UO_2(NO_3)_4^2 + 2NO_3^-$$

 $2R-NO_3^- + Th(NO_3)_6^2 + R_2-Th(NO_3)_6^2 + 2NO_3^-$

Concentration:

$$R_2$$
-UO₂(NO₃)₄²⁻ + 2Cl⁻ $= 2R$ -Cl⁻ + UO₂²⁺ + 4NO₃⁻
 R_2 -Th(NO₃)₆²⁻ + 6Cl⁻ $= 2R$ -Cl⁻ + ThCl_{4(aq)} + 6NO₃⁻

Quantitative Analysis

- 1. a linear external calibration
- the recovery of standard addition | Recovery = $\frac{A_{tracer}}{c} \times 100\%$



certified for		
IRMM	051	/231
2.4 mg ^{2.8} U/kg	²³³ U/U=0.98	m(²³⁴ m(²³⁴ m(²³⁶ m(²³⁶
		m(^{23/}

mass fraction (·100) U)/m(U)98.020 1(58) U)/m(U)0.918 3(02) U)/m(U)0.216 0(56) U)/m(U)0.024 4(03) U)/m(U)0.821 2(20)

2.3539(42)ppm ²³³U



mass fraction (-100) $m(^{230}Th)/m(Th)$ 99.849(50) $m(^{232}Th)/m(Th)$ 0.151(50)

0.5691(42)ppm ²³⁰Th



mass fraction (·100) $m(^{234}U)/m(U)$ 0.035 9(02) $m(^{235}U)/m(U)$ 99.941 4(18) $m(^{236}U)/m(U)$ 0.009 6(10) $m(^{238}U)/m(U)$ 0.013 1(03)

0.9999(25)ppm ²³⁵U



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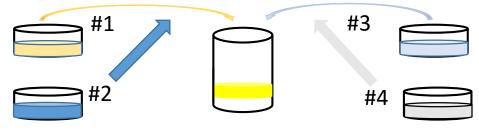
Experimental Section

1. Cu



flow rate: 0.05ml/min

Pre-cleaning:



#1: 2 ml 8M HNO₃ #2: 2ml UPW

#3: 2ml HCl (37%) #4: 8ml 0.1 M HCl

Concentration:

$$2R-NO_{3}^{-} + UO_{2}(NO_{3})_{4}^{2} + 2R_{2}-UO_{2}(NO_{3})_{4}^{2-} + 2NO_{3}^{-}$$

$$2R-NO_{3}^{-} + Th(NO_{3})_{6}^{2-} + R_{2}-Th(NO_{3})_{6}^{2-} + 2NO_{3}^{-}$$

$$R_{2}-UO_{2}(NO_{3})_{4}^{2-} + 2Cl^{-} + 2R-Cl^{-} + UO_{2}^{2+} + 4NO_{3}^{-}$$

$$R_{2}-Th(NO_{3})_{6}^{2-} + 6Cl^{-} + 2R-Cl^{-} + ThCl_{4(aq)} + 6NO_{3}^{-}$$

Sample 8 M HNO₃ loaded HNO₃

0.1 M

HCI

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Results and discussion

Initial Cu sample (g): 9.05589

Surface preparation (g): 6.26617

Amount of digestion (g): 6.26617

1.1 20160615

Recovery = $\frac{A_{tracer}}{C_{tracer}} \times 100\%$

7	Ī	ľ		3	N	C	1	2	1	٢		2	3	}	5	l				ſ		N	0	r	١	٠	• :		٠	•	Ī	٦	•		M					- 2	2	3	5	i		1		. 1	1			P		_	N	V	1	S		
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621.136 1.1922(0.0155)

²³²Th_ana (pg) Tot-in-Cu (pg)

5.9547(0.1664) 2933.7286

²³⁸U_ana (pg) Tot-in-Cu (pg)

1.1966(0.0058) 560.5987

Recovery

0.1919%

Concentration(pg/g)

468.1853(0.1664)

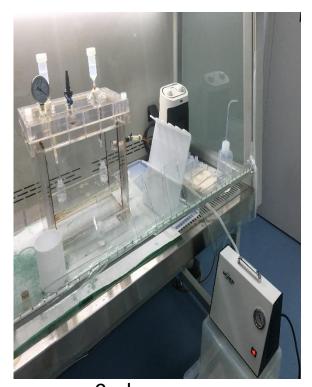
Concentration(pg/g)

89.4643(0.0058)

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Experimental Section

2. A comparison of two composite resins

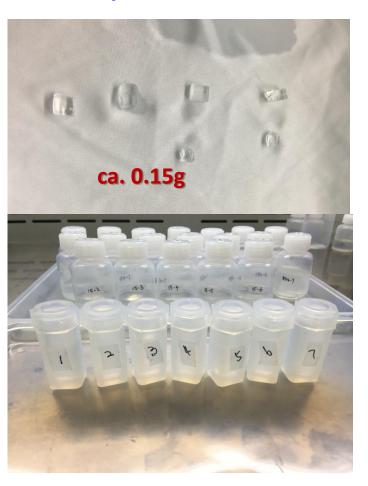


ca. 9ml flow rate: 1 ml/min

Sample	232 Th CPS (RSD)	235 U CPS (RSD)	238 U CPS (RSD)
Watson water	1	-0.27	1.733
	(20)	(-128.300)	(24.019)
Watson water resin-I	0.867	-0.27	-0.067
	(13.323)	(-74.074)	(-346.410)
Watson water resin-D	0.733	-0.403	0.133
	(56.773)	(-28.629)	(312.250)
Spiked solution ca. 0.3 pg/g	343.607	503.673	424.01
	(0.717)	(1.949)	(2.795)
Spiked solution resin-I	2.067	-0.27	2.133
	(29.565)	(-128.300)	(48.109)
Spiked solution resin-D	2.533	-0.137	1.133
	(31.906)	(-223.540)	(100.345)

Experimental Section

3. Acrylic



the problems:

> Efficient cutting method

> The total chemical blank

➤ Fast cleaning for high pressure digestion tank

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Summary

- 1. Quickly and convenient cleaning technology for all the vials, containers and lids
- 2. The recovery of standard addition
- 3. Minimize the total chemical blank: ✓ vessels, reagent, specification operation
- 4. Seek optimal experimental scheme and equipment

Still need learn more about trace analysis!

Thanks

Any suggestion or question will be thankful!

袁影

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